

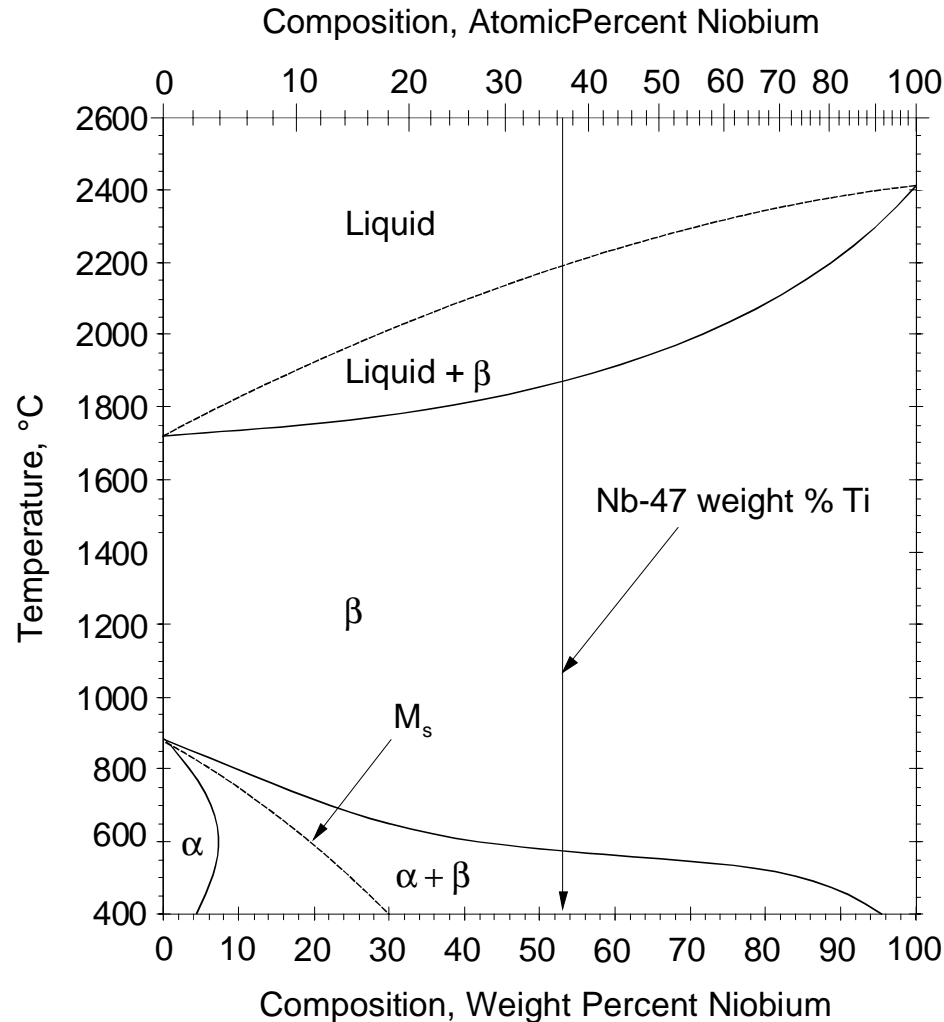


## II. Basic Materials Issues

- Crystal Structures
  - Nb-Ti: body centered cubic *simple, ductile*
  - Nb<sub>3</sub>Sn: cubic A15 Crystal structure with range of off-stoichiometric compositions
  - BSCCO: complex layered phase(s) not found at fixed stoichiometry of nominal phase
  - YBCO: layered phase of fixed cation stoichiometry but variable O content
- Only Nb-Ti is ductile
- Essential Phase Diagram Information
  - well known for LTS, poorly known for HTS



# Ila. Nb-Ti Phase Diagram

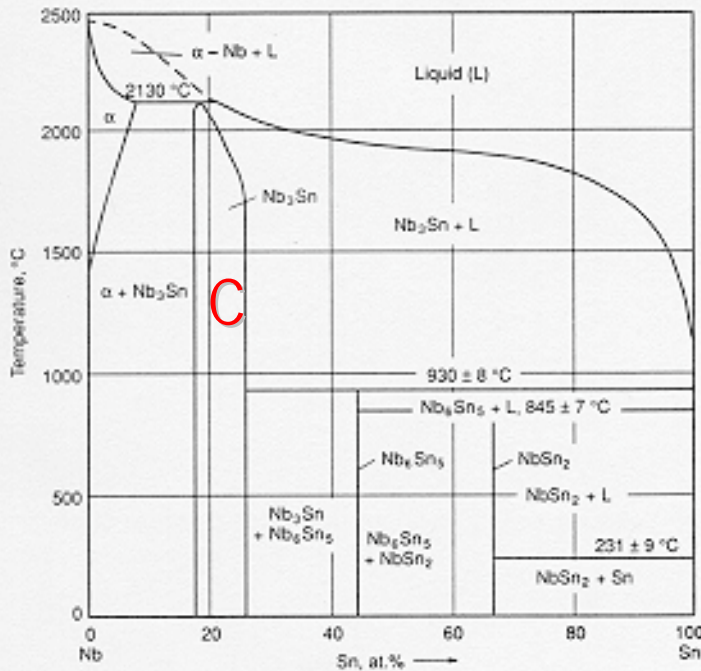


## Key features:

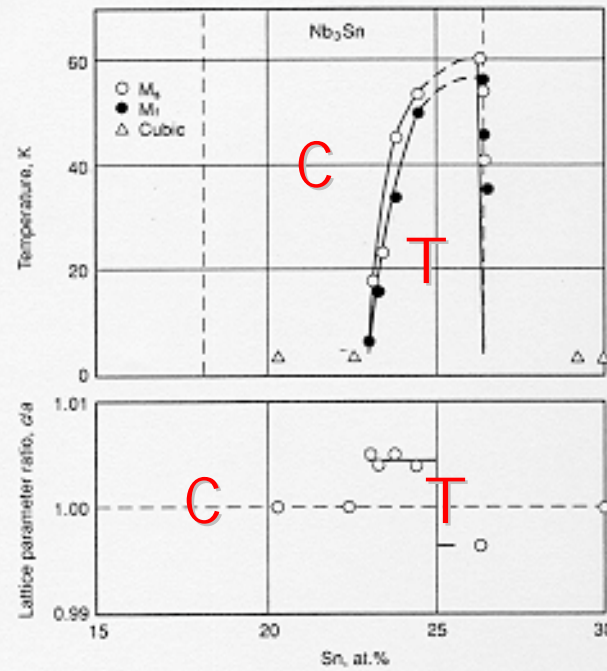
- Very high melting points
- Large separation liquidus and solidus leads to segregation on cooling
- Nb is bcc at all T, while Ti is bcc at high T and hcp at low T
- Nb-Ti alloys want to become 2 phase hcp and bcc at low T but cannot transform by diffusion
- The lattice acquires a soft phonon that has very important consequences:
  - E declines on cooling
  - $\rho$  increases on cooling
- The martensitic phase transformation is only incipient for Nb47wt.%Ti but the resistivity is greatly increased and  $H_{c2}$  increased too



# IIb. Nb-Sn Phase Relations



(a)



(b)

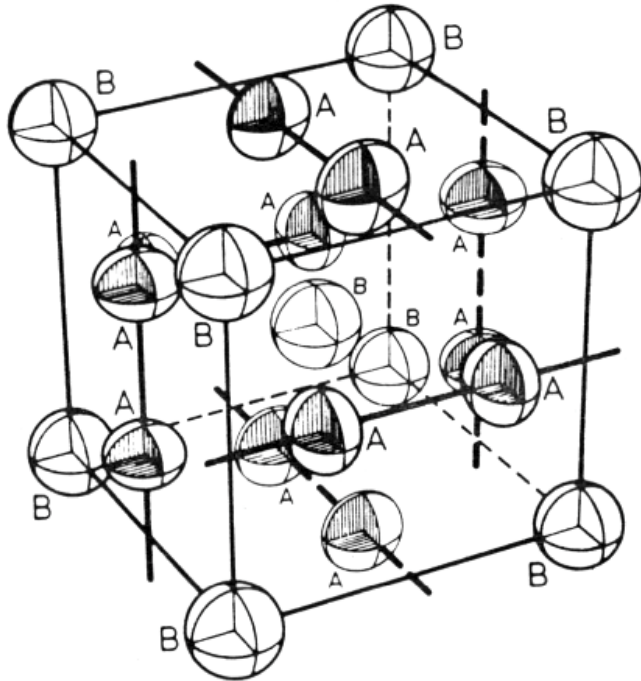
Fig. 12 Niobium-tin binary phase diagram. (a) Elevated temperatures. (b) Subzero temperatures.  $M_f$ , temperature at which martensite formation finishes during cooling;  $M_s$ , temperature at which martensite starts to form on cooling. Sources: Ref 11, 12

C: Cubic A15, higher  $H_{c2}$  phase, T: Tetragonal A15 phase

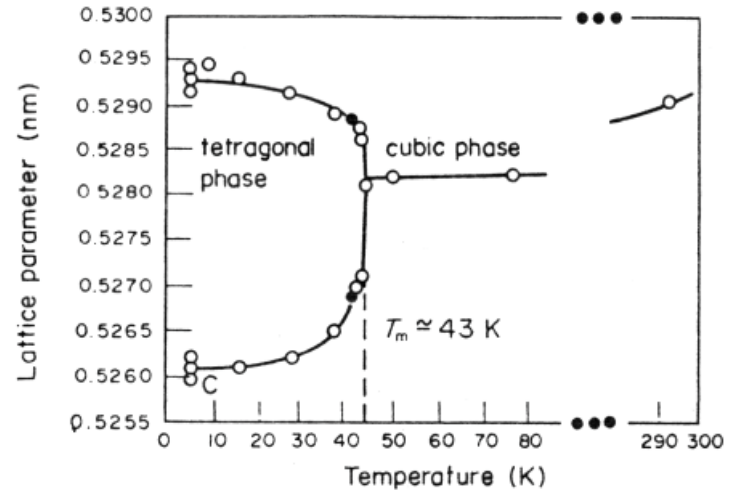
Broad composition range 18-25at.%Sn, LT shear transformation to a small tetragonality, lowering  $T_c$  and  $H_{c2}$



# ***IIc. Nb<sub>3</sub>Sn Structure***



**Figure 1**  
The A15 type structure A<sub>3</sub>B, space group *Pm3n*



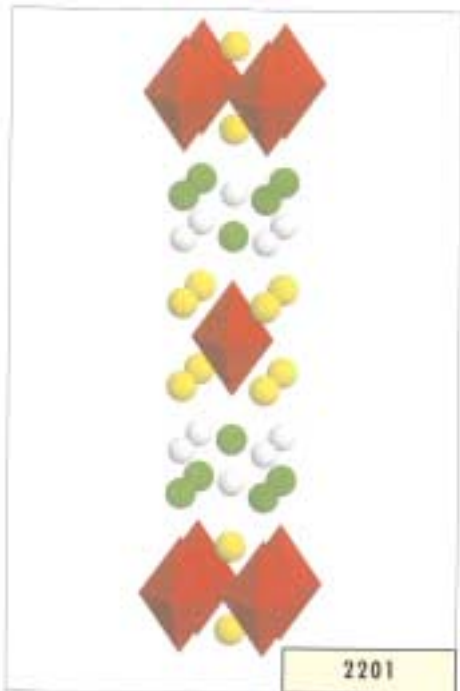
**Figure 2**  
Structural instability in Nb<sub>3</sub>Sn

Cubic structure with 3 orthogonal chains in which the Nb atoms are more closely spaced than in pure Nb: high N(0).

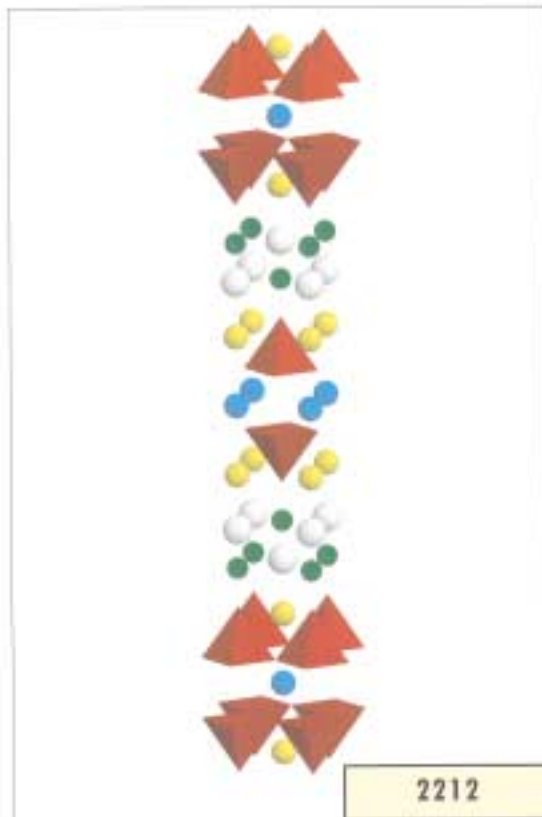
Departures from stoichiometry must be accommodated by vacancies or anti-site disorder



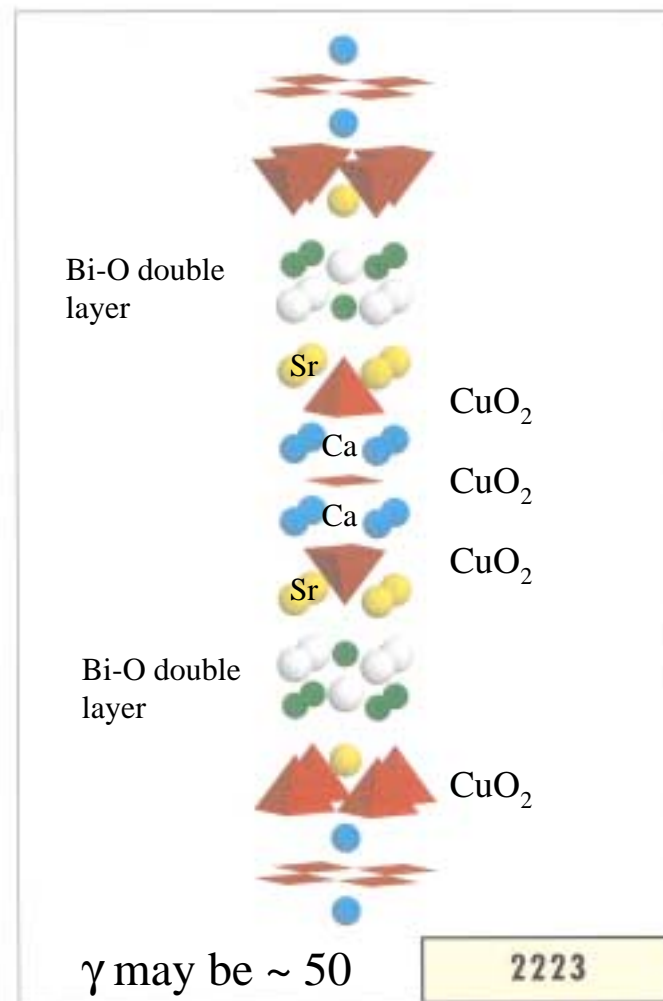
# IId. The BSCCO Family



$T_c \sim 30\text{K}$



$\gamma \sim 200-300$   
 $\sim 70-95\text{K}$

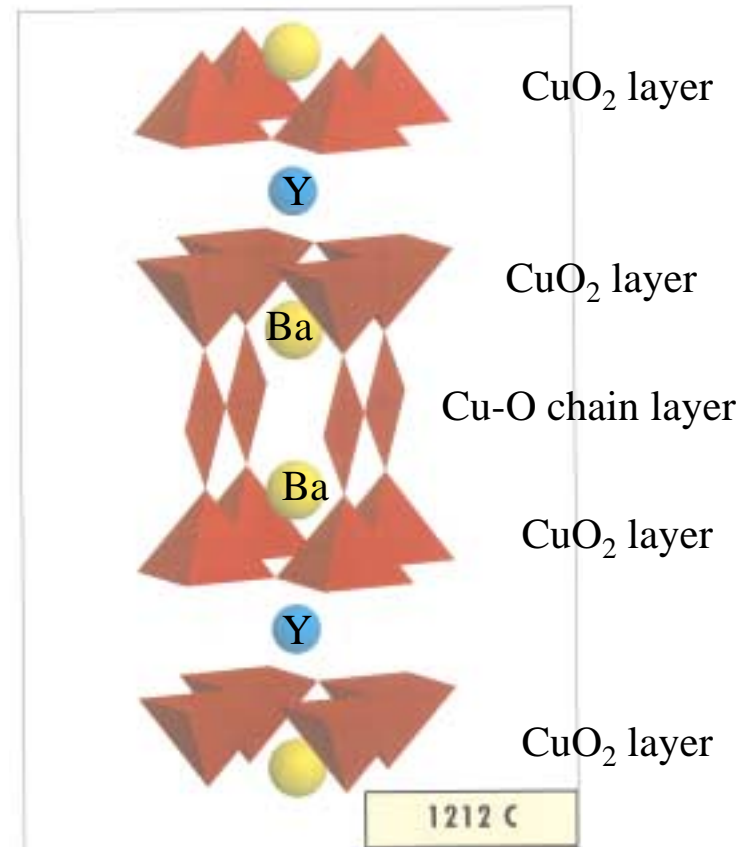


$\gamma$  may be  $\sim 50$   
 $\sim 105-110\text{K}$



## Ile. YBCO

- YBCO ( $\text{YBa}_2\text{Cu}_3\text{O}_{7-x}$ ) possesses the first crystal structure with  $T_c > 77\text{K}$
- It has defined cation stoichiometry 1:2:3, but O content is variable from 6-7
  - $T_c > 90\text{K}$  demands  $x < 0.05$
- YBCO is often thought of as being archetypal but in fact the Cu-O chain layers are very unusual
  - Make charge reservoir layer metallic
  - Most HTS are 2D, but YBCO is anisotropic 3D with electron mass anisotropy  $\gamma = (m_c/m_a)^{0.5}$  of  $\sim 7$





# ***Ilf Summary Materials Issues***

- Higher  $T_c$  means greater crystal complexity
  - body centered cubic Nb-Ti with random site occupation, **ductile**
    - phonon anomalies
  - Nb<sub>3</sub>Sn ordered A15 phase, **brittle**
    - other intermetallics often distort phase field
  - YBCO is anisotropic 3D, low carrier density, cation stoichiometric, **brittle**
    - metallic charge reservoir layer
  - BSCCO has 3 layered phases none of which exist at cation stoichiometry, **micaceous with self-aligning tendencies**
    - strongly anisotropic, 2D (but depends on doping state)
    - poorly understood phase relations
- **Materials quality at scale of  $\xi$  is always an issue!**