

# Stochastic thermodynamics

Udo Seifert

II. Institut für Theoretische Physik, Universität Stuttgart

thanks to:

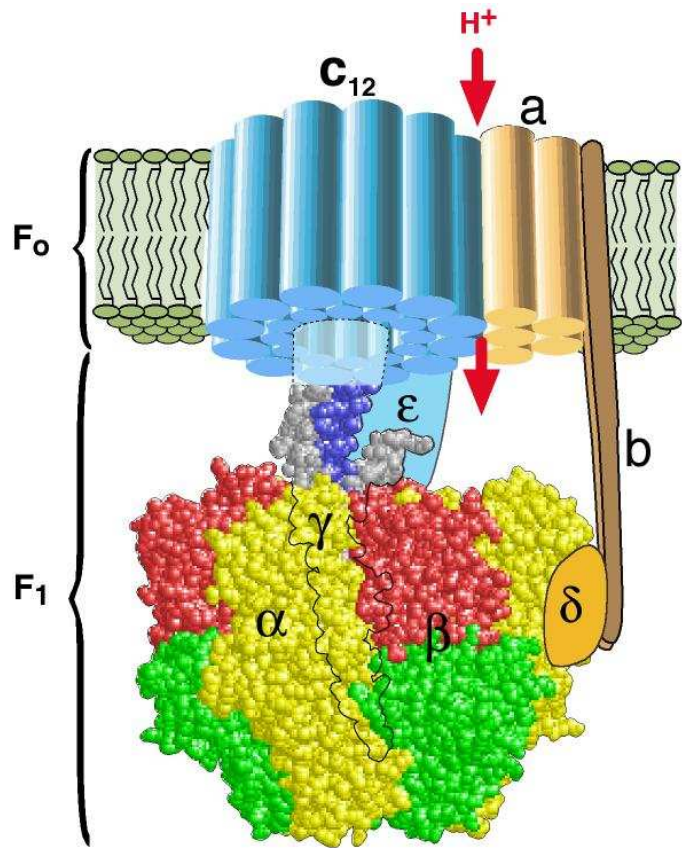
- F. Berger, J. Mehl, T. Schmiedl and Th. Speck (theory)
- V. Blickle and C. Bechinger (expt's on colloidal particles)
- C. Tietz, S. Schuler and J. Wrachtrup (expt's on single atoms)

Review: U.S., Eur. Phys. J. B, 64 : 423-431, 2008.

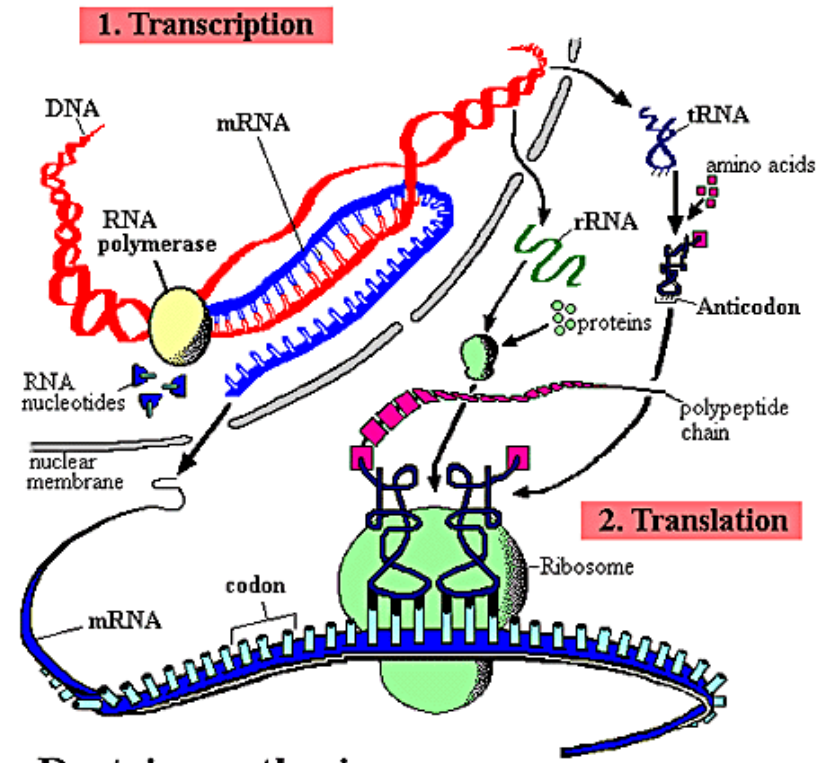
## LECTURE III

- Stochastic dynamics along discrete states
- Fluctuation theorems with examples
- Biochemically driven systems

- (Bio)chemically driven systems



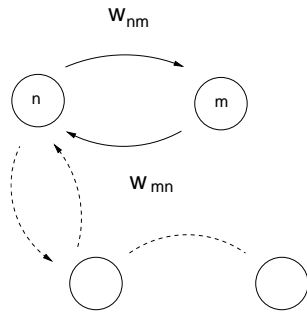
H. Wang and G. Oster (1998). Nature 396:279-282.



Protein synthesis

F1-ATPase

- Master equation dynamics on discrete states



states  $\{n\}$  could be

- internal states of a protein
- number of a species
- discrete spatial coordinate
- ....

- time-dependent rates given by a “protocol”  $\lambda(\tau)$

- $\partial_\tau p_n = \sum_m [w_{mn}(\lambda)p_m - w_{nm}(\lambda)p_n]$

- solution  $p_n(\tau)$  depends on initial  $p_n(0)$

- unique stationary solution  $p_n^s(\lambda)$  for any fixed  $\lambda$

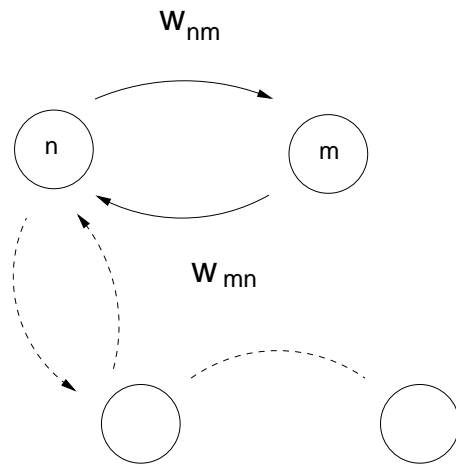
- \* detailed balanced

$$p_n^s w_{nm} = p_m^s w_{mn} \rightarrow \frac{w_{nm}}{w_{mn}} = \exp(E_n - E_m) \text{ with } E_n \equiv -\ln p_n^s$$

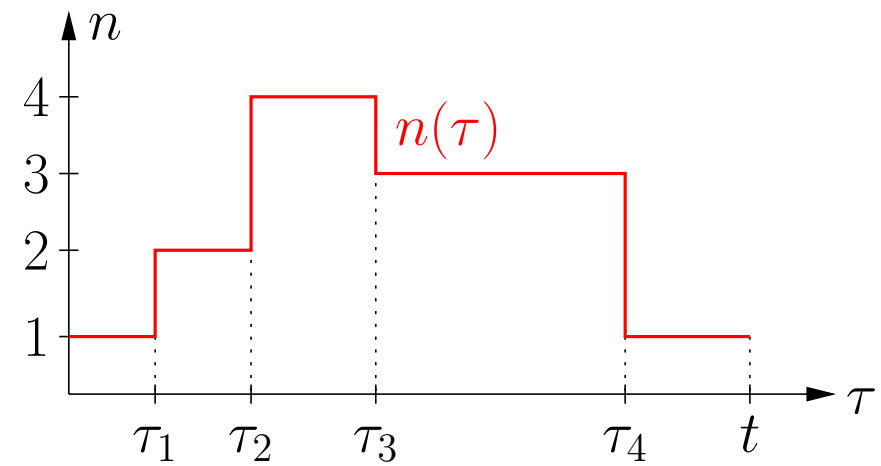
- \* persistent cycle currents

$$p_n^s w_{nm} \neq p_m^s w_{mn}$$

- Stochastic trajectory



– jumps at  $\tau_j$  from  $n_j^-$  to  $n_j^+$



## Stochastic entropy [U.S., PRL **95**, 040602 (2005)]

- Non-equilibrium ensemble entropy

$$S(\tau) \equiv - \sum_n p_n(\tau) \ln p_n(\tau) = \langle s(\tau) \rangle$$

- Stochastic (trajectory-dependent) entropy of the system

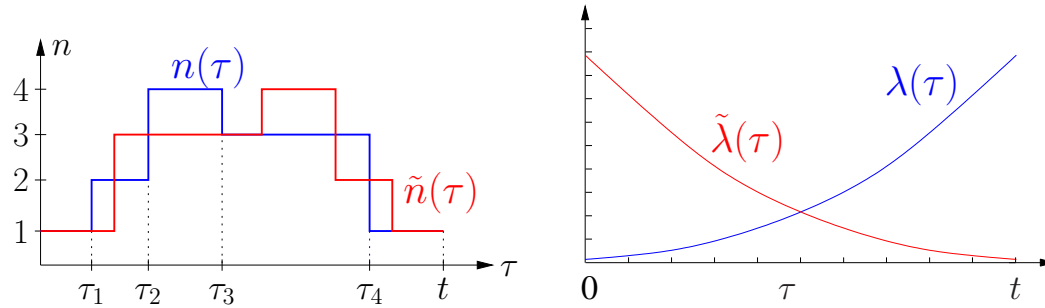
$$s(\tau) \equiv - \ln p_{n(\tau)}(\tau)$$

- equation of motion

$$\begin{aligned} \dot{s}(\tau) &= \underbrace{-\frac{\partial_\tau p_n(\tau)}{p_n(\tau)} \Big|_{n(\tau)} - \sum_j \delta(\tau - \tau_j) \ln \frac{p_{n_j^+}(\tau_j)}{p_{n_j^-}(\tau_j)}}_{\equiv \dot{s}_{\text{tot}}(\tau)} + \underbrace{\sum_j \delta(\tau - \tau_j) \ln \frac{w_{n_j^+ n_j^-}}{w_{n_j^- n_j^+}}}_{\equiv -\dot{s}_m(\tau)} \\ &= \underbrace{-\frac{\partial_\tau p_n(\tau)}{p_n(\tau)} \Big|_{n(\tau)} - \sum_j \delta(\tau - \tau_j) \ln \frac{p_{n_j^+} w_{n_j^+ n_j^-}}{p_{n_j^-} w_{n_j^- n_j^+}}}_{\equiv \dot{s}_{\text{tot}}(\tau)} + \sum_j \delta(\tau - \tau_j) \ln \frac{w_{n_j^+ n_j^-}}{w_{n_j^- n_j^+}} \end{aligned}$$

- $\langle \dot{s}_{\text{tot}}(\tau) \rangle \geq 0$

- “Time reversal”



$$\tilde{n}(\tau) \equiv n(t - \tau) \text{ and } \tilde{\lambda}(\tau) \equiv \lambda(t - \tau)$$

- Ratio of backward to forward path

$$\frac{\tilde{p}[\tilde{n}(\tau)|\tilde{n}_0]}{p[n(\tau)|n_0]} = \exp[-\Delta s_m]$$

- general fluctuation theorem

$$\begin{aligned}
 1 &= \sum_{\tilde{n}(\tau), \tilde{n}_0} \tilde{p}[\tilde{n}(\tau)|\tilde{n}_0] p_1(\tilde{n}_0) \\
 &= \sum_{n(\tau), n_0} p[n(\tau)|n_0] p_0(n_0) \frac{\tilde{p}[\tilde{n}(\tau)|\tilde{n}_0] p_1(\tilde{n}_0)}{p[n(\tau)|n_0] p_0(n_0)} \\
 &= \langle \exp[-\Delta s_m + \ln p_1(n_t)/p_0(n_0)] \rangle
 \end{aligned}$$

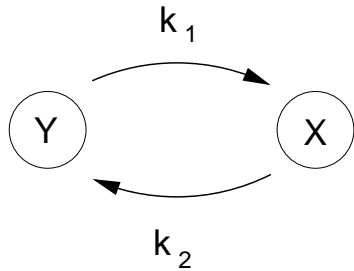
- choice 1:  $p_1(n) = p(n, t) \Rightarrow \langle \exp[-\Delta s_{\text{tot}}] \rangle = 1 \Rightarrow \langle \Delta s_{\text{tot}} \rangle \geq 0$   
 – arbitrary initial distribution, arbitrary driving, finite times
- choice 2:  $p_1(n) = p^s(n, \lambda_t) \Rightarrow \langle \exp(-R) \rangle = 1$   
 – with  $R = -\int_0^t \partial_\tau \ln p_n^s(\lambda(\tau))|_{n(\tau)}$   
 – athermal discrete generalization of the Jarzynski relation
- similarly for a NESS: DFT  $p(-\Delta s_{\text{tot}})/p(\Delta s_{\text{tot}}) = \exp(-\Delta s_{\text{tot}})$



# Illustration: Birth-death or chemical master equations

[U.S., J Phys A 37, L517, 2004]

- Simplest case: Isomerization



–  $n_X \equiv n = N - n_Y$

–  $q(\tau) \equiv k_1(\tau)/k_2(\tau)$

– stationary distribution:  $p^s(n) = [q/(1 + q)]^N \binom{N}{n}$

– stationary mean  $n^s = Nq/(1 + q)$

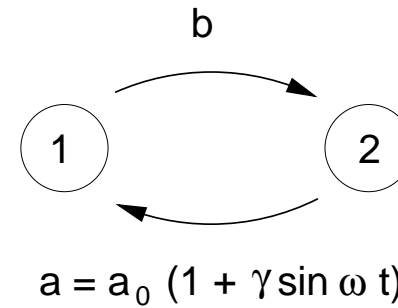
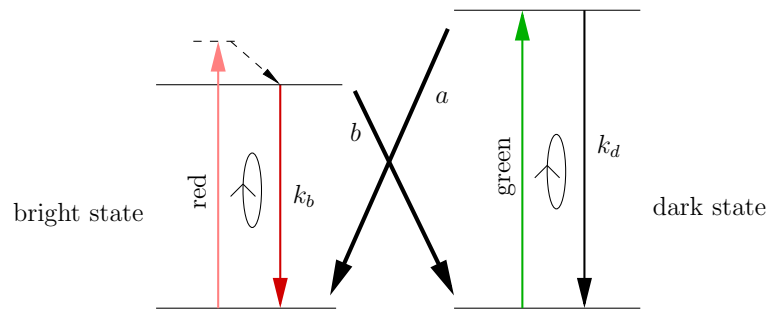
–  $\langle \exp\{ \int_0^t d\tau \underbrace{[n(\tau) - n^s(\tau)]}_{\text{non-eq lag}} \frac{d}{d\tau} \ln q(\tau) \} \rangle = 1$

– follow-up: C. Jarzynski, J. Phys. A. 38, L227, 2005

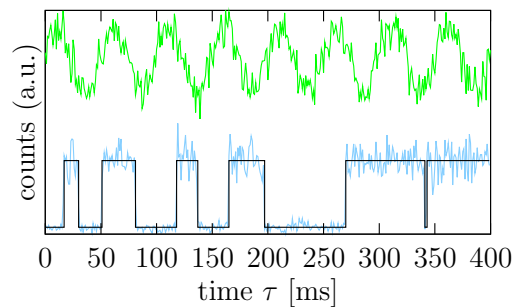
# Periodically driven system: Optically active defect center in diamond

[S. Schuler, T. Speck, C. Tietz, J. Wrachtrup and U.S., PRL **94**, 180602, 2005

C. Tietz, S. Schuler, T. Speck, U. S., and J. Wrachtrup, PRL **97** 050602, 2006]

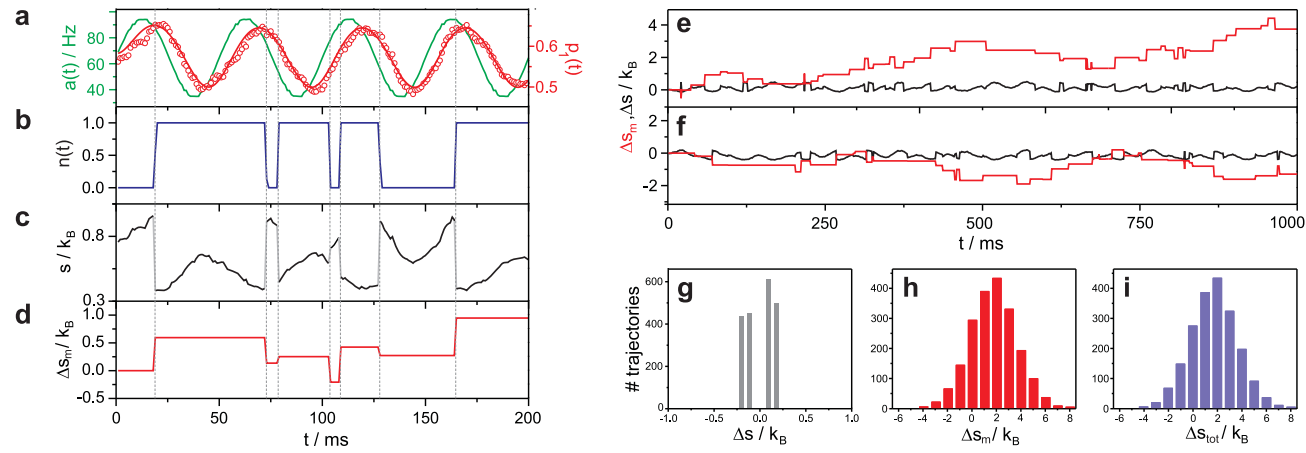


- Trajectories



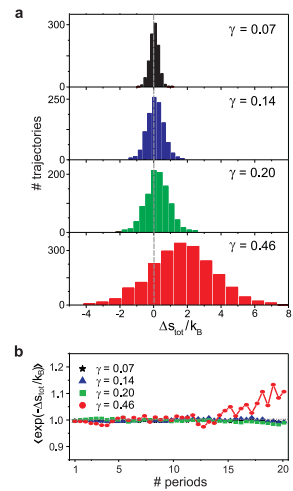
- Entropy production along a single trajectory

Figure 1



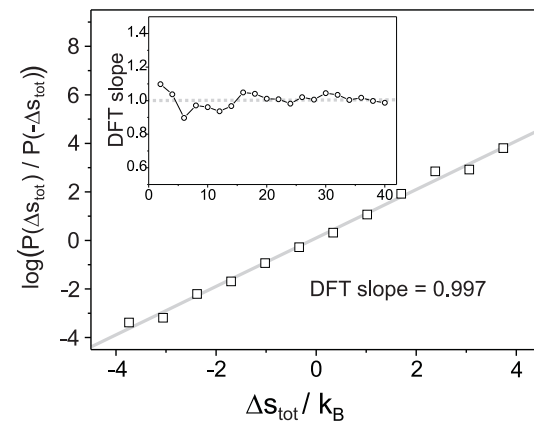
- Int FT

Figure 2



- Det FT

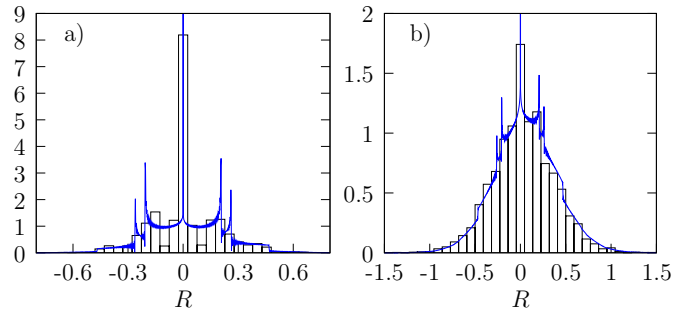
Figure 3



- Generalized JR:

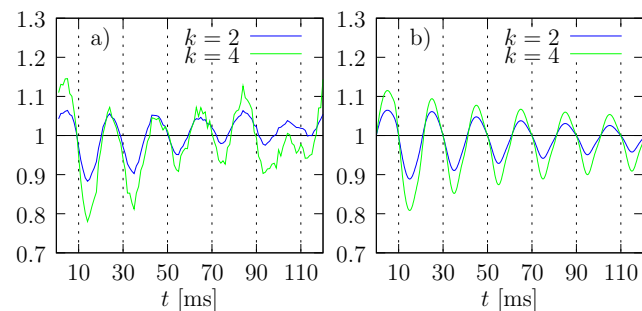
$$\langle \exp[-R] \rangle = 1 \quad \text{for} \quad R[n(\tau)] \equiv - \int_0^t d\tau \lambda \partial_\lambda \ln p_{n(\tau)}^s(\lambda) \quad (= W - \Delta F)$$

probability distribution  $p(R)$



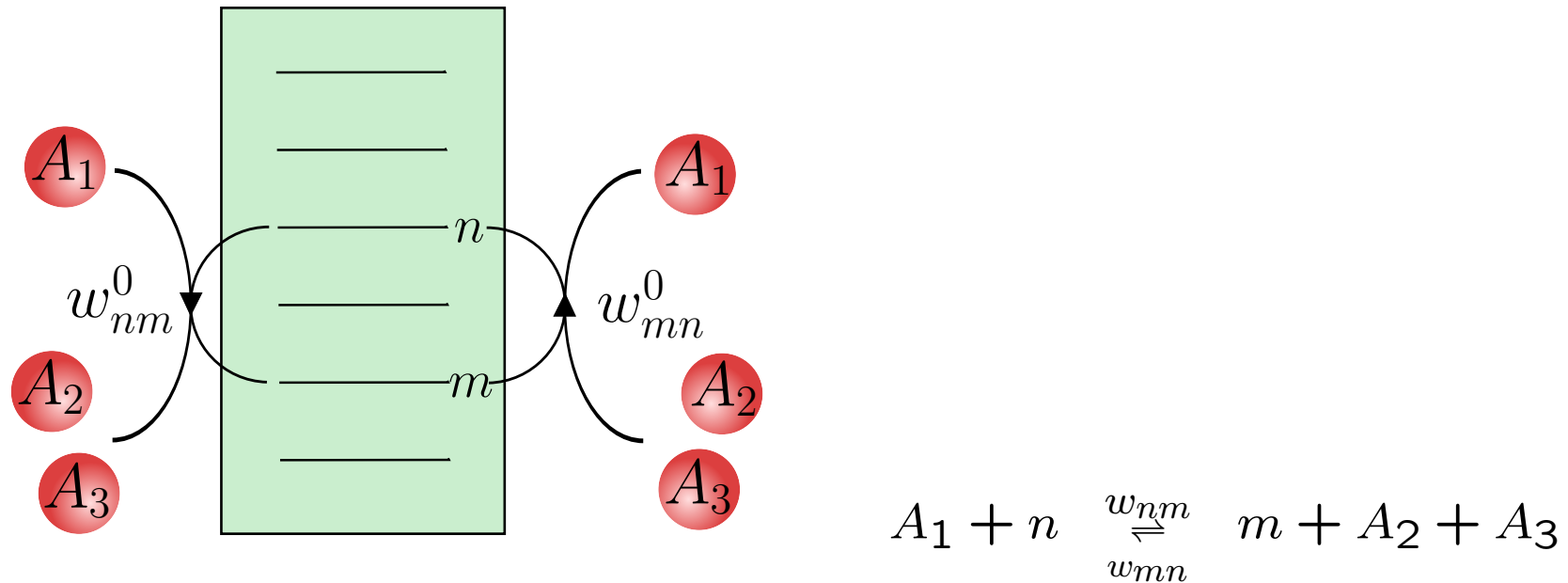
- Detailed theorem for symmetric protocols  $\lambda(\tau) = \lambda(t - \tau)$ :

$$p(-R)/p(R) = \exp(-R) \quad \Rightarrow \quad \langle R^k \rangle = (-1)^k \langle R^k \exp(-R) \rangle$$



- Stochastic th'dynamics of a driven enzyme with internal states

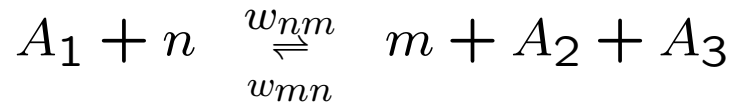
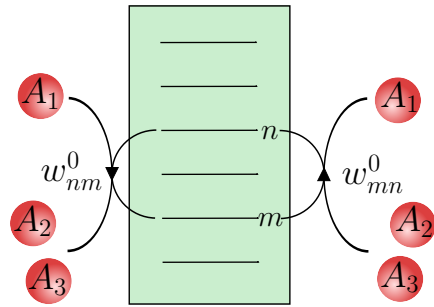
[T.Schmiedl, T.Speck and U.S., J. Stat. Phys. **128**, 77 (2007)]



- Green enzym
  - \* internal states with energy  $E_n$
- Red species
  - \* fixed (non-equilibrium) chemical potential

$$\mu = \mu_\alpha^0 + \ln \{ [A_\alpha] / c^0 \} = E_\alpha + \ln \{ [A_\alpha] \omega_\alpha \}$$

- Mass action law kinetics



$$- \frac{w_{nm}}{w_{mn}} = \frac{w_{nm}^0}{w_{mn}^0} [A_1] / [A_2][A_3]$$

$$- \text{in hypothetical eq.: } \frac{w_{nm}^{\text{eq}}}{w_{mn}^{\text{eq}}} = \frac{w_{nm}^0}{w_{mn}^0} \{ [A_1] / [A_2][A_3] \}^{\text{eq}} = \frac{p_m^{\text{eq}}}{p_n^{\text{eq}}} = \exp(-\Delta G)$$

with

$$\Delta G \equiv -[E_n - E_m + \{\mu_1 - \mu_2 - \mu_3\}^{\text{eq}}]$$

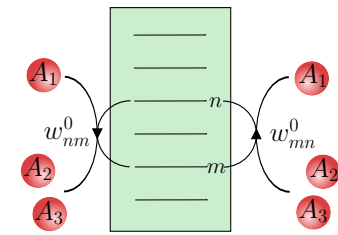
and

$$\mu_\alpha^{\text{eq}} \equiv E_\alpha + \ln \{ [A_\alpha]^{\text{eq}} \omega_\alpha \}$$

$$- \text{Ratio of intrinsic rates: } \frac{w_{nm}^0}{w_{mn}^0} = \frac{\omega_1}{\omega_2 \omega_3} \exp[E_n - E_m + (E_1 - E_2 - E_3)]$$

relation between the bare rates and the energy is indep't of  $[A_\alpha]$

- First law along stochastic trajectory



$$w = \Delta E + q \quad \text{for a single step}$$

- chemical work:  $w_{\text{chem}}^{nm} \equiv \mu_1 - \mu_2 - \mu_3$

- internal energy:  $\Delta E^{nm} \equiv E_m - E_n$

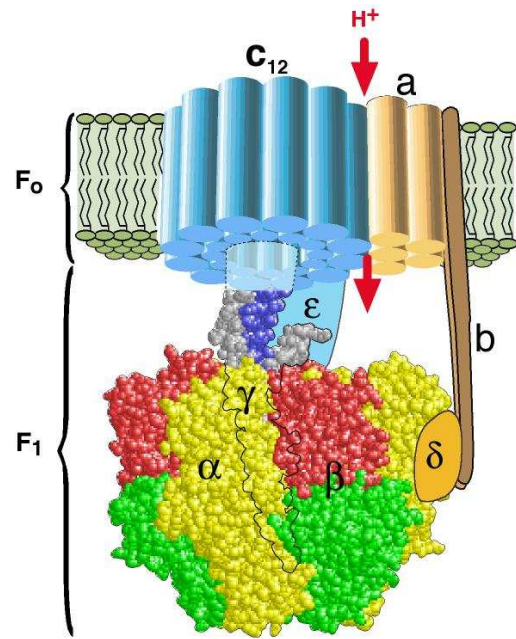
- dissipated heat:  $q^{nm} \equiv w_{\text{chem}}^{nm} - \Delta E^{nm} = \ln \frac{[A_1]}{[A_2][A_3]} \frac{w_{nm}^0}{w_{mn}^0}$

- previous abstract definition:  $\Delta s_m^{nm} \equiv \ln w_{nm}/w_{mn} = q^{nm}$

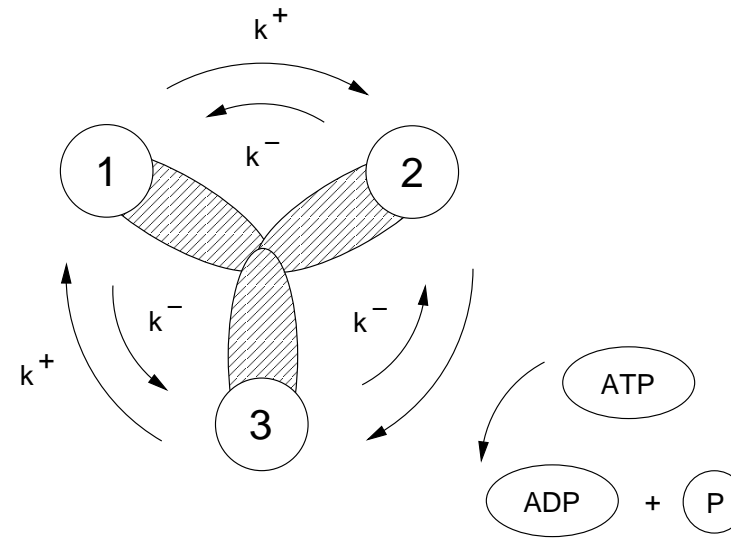
- abstract approach consistent with usual phys chem

- can be summed over full reaction history

● Illustration: F<sub>1</sub>-ATPase [U.S., Europhys. Lett. 70, 36, 2005]



H. Wang and G. Oster (1998). Nature 396:279-282.



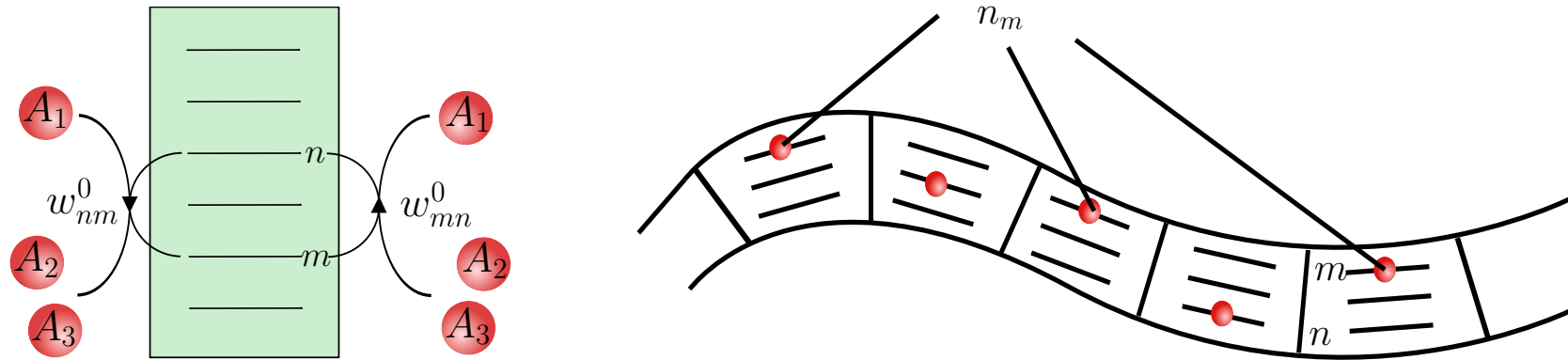
$$- \partial_{\tau} p_1 = -(k^+ + k^-) p_1 + k^+ p_2 + k^- p_3 \quad \& \quad \text{cyc}$$

$$- \Delta s_{\text{tot}} = n \ln(k^+ / k^-) = n [\mu_{\text{ATP}} - \mu_{\text{ADP}} - \mu_{\text{P}}] / T$$

$$- \boxed{p(-n) / p(n) = \exp[-n \ln(k^+ / k^-)]}$$



- Degeneracy



$N$  of those                      OR                      N-domain protein/...

–  $\mathbf{n}(\tau) = (n_1(\tau), n_2(\tau), \dots, n_M(\tau))$

– for  $n_n \rightarrow n_n - 1, \quad n_m \rightarrow n_m + 1$

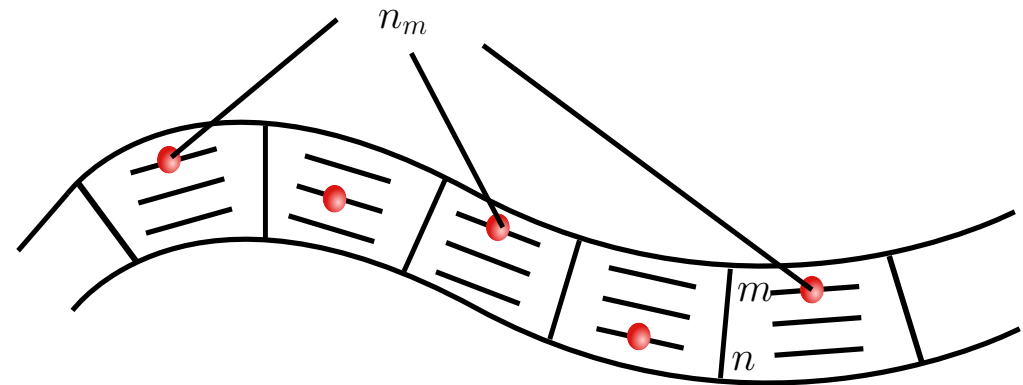
$$\frac{W_{nm}}{W_{mn}} = \frac{w_{nm}n_n}{w_{mn}(n_m+1)}$$

– heat released  $\Delta s_m^{nm} = \ln \frac{w_{nm}}{w_{mn}}$  as before

– but:  $\Delta s_m^{nm} = \ln \frac{W_{nm}}{W_{mn}} = \ln \frac{w_{nm}}{w_{mn}} + \ln [n_n / (n_m + 1)]$

– an apparent inconsistency?

- Degeneracy cont'd



- for  $n_n \rightarrow n_n - 1$ ,  $n_m \rightarrow n_m + 1$

$$\frac{W_{nm}}{W_{mn}} = \frac{w_{nm}n_n}{w_{mn}(n_m+1)}$$

- heat released  $\Delta s_m^{nm} = \ln \frac{w_{nm}}{w_{mn}}$  as before

- but:  $\Delta s_m^{nm} = \ln \frac{W_{nm}}{W_{mn}} = \ln \frac{w_{nm}}{w_{mn}} + \ln [n_n / (n_m + 1)]$

- solution

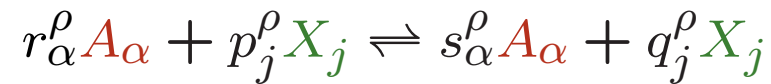
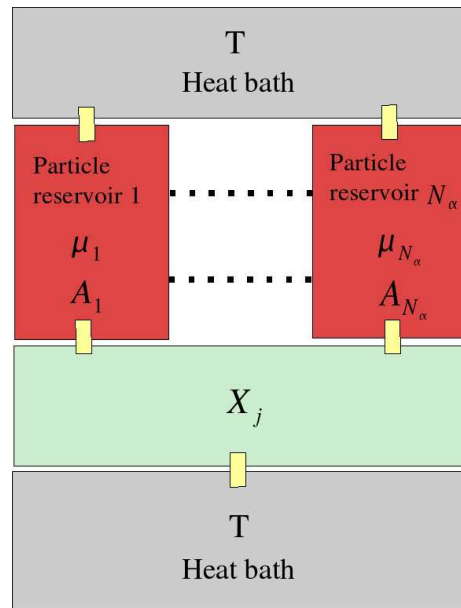
- $s^n = s_0^n + \ln g^n$

- $\Delta s_m^{nn'} \equiv \Delta s_{m0}^{nn'} - \ln(g^n / g^{n'})$

- FT's for  $\Delta s_{\text{tot}}$  remain valid

- Stochastic th'dynamics of general (bio)chemical reaction networks

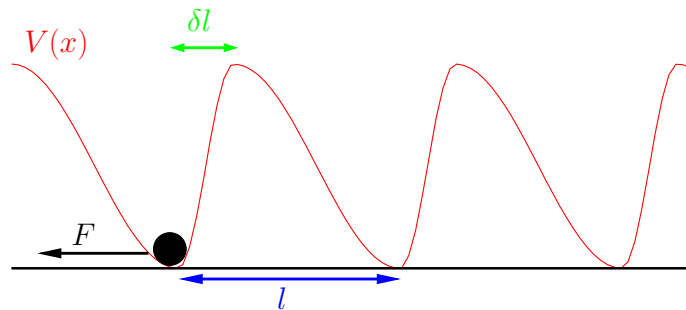
[ T. Schmiedl and U.S., J. Chem. Phys. 126, 044101, 2007]



- two sorts of species:
  - \* chemiostated  $A_{\alpha}$  (like ATP, ADP, ...)
  - \* number-tracked  $X_j$  with  $n_j(\tau)$
- energy and entropy along a trajectory  $\{n_j(\tau)\}$

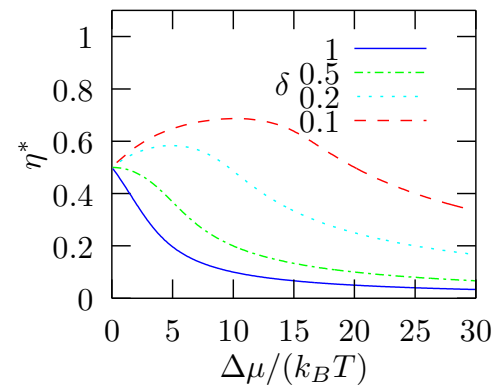
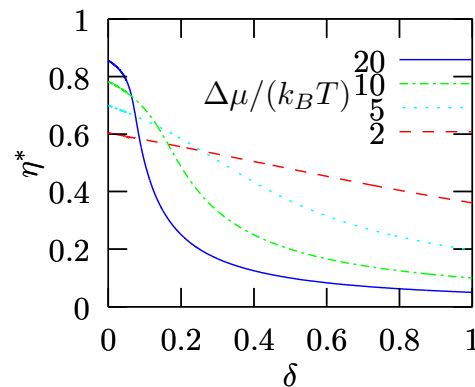
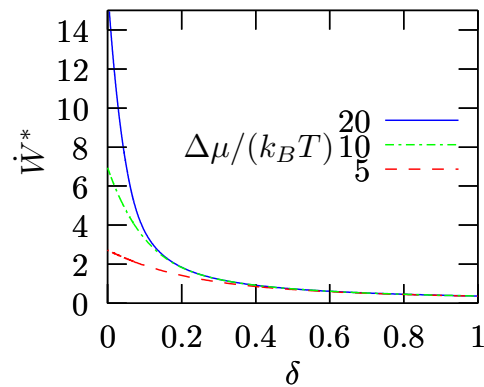
- Efficiency of molecular motors at maximum power

[T. Schmiedl and U.S., EPL 83: 30005, 2008]



$$w^+ = [ATP]k^+ \exp[-\delta l F]$$

$$w^- = [ADP][P]k^- \exp[(1 - \delta)l F]$$



- “Power stroke” ( $\delta \simeq 0$ ) highest efficiency at max power
- $\eta^*$  can increase beyond lin response regime ( $\eta^* = 1/2$ )

- **Stochastic thermodynamics** along single trajectories

- formulation of the 1<sup>st</sup> law

- refinement of the 2<sup>nd</sup> law

- generalized FDT's

- \* mechanically or flow driven: colloids, polymers, proteins

- \* biochemically driven: single enzymes, motors, reaction networks

- \* optically driven: defect centers, other quantum systems ....

- generalizations

- \* many interacting degrees of freedom (conc. done, pract. open)

- \* stochastic field theories (like KPZ-eq)

- \* quantum systems (JR done, DFT open)

- \* ...