

Lecture 1: Neutral brushes. Scaling model of a neutral planar polymer brush (mushroom and brush regimes). Effect of solvent. Strong stretching approximation: chain trajectory and parabolic potential. Internal structure of a planar brush. Response of polymer brush to compression. Curved polymer brushes. Scaling model of star-like and comb-like molecular brushes (stars and combs in solution).

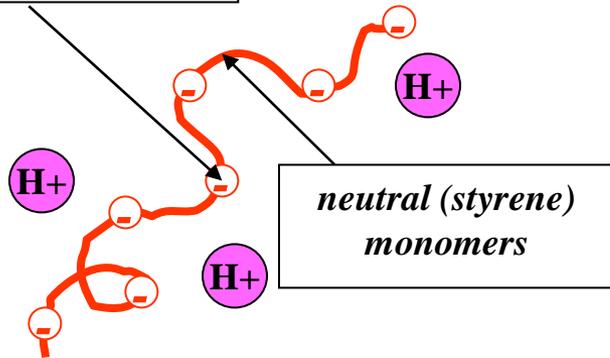


Lecture 2: Charged brushes. Strong and weak polyelectrolytes. Scaling model of strong PE planar brush. Main regimes of PE brush (counterion and salt dominated). Local electroneutrality approximation. Parabolic potential and internal structure of planar PE brush. Interactions between planar PE brushes. Curved PE brushes (scaling model).  
**Corona of neurofilament (NF) as a cylindrical PE brush.**

# Strongly and weakly dissociating polyelectrolytes

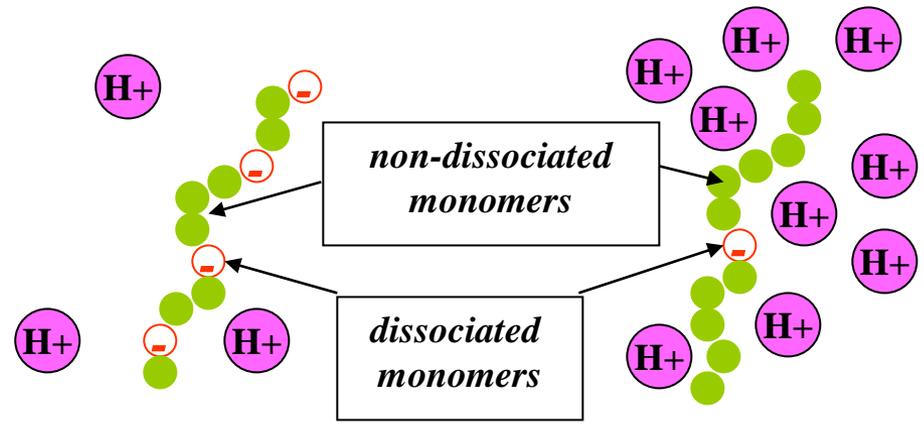
## Strong polyelectrolyte:

*permanently charged  
(sulfonated) monomer*



***Poly(styrene/styrenesulfonate)**  
fraction of charged monomers is  
independent of salinity, pH, ...,  
pre-determined by chemical  
composition (sequence)*

## Weak polyelectrolyte (polyacid):



*high pH*

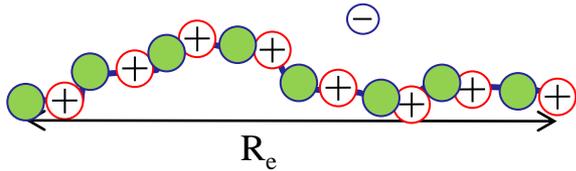
*low pH*

***Poly(acrylic acid)***

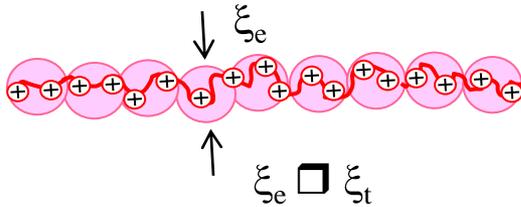
*fraction of charged monomers  
(=degree of ionization) depends on local  
concentration of  $H^+$  ions ( $pH = -\log[H^+]$ )  
via ionization/recombination equilibrium*

# Individual polyion and charged surface

Theta solvent for flexible backbone  
 Net charge per chain  $Q = fN$



Electrostatic blob = tension blob



Polyion size  $R_e$  is governed by balance of  
 Gaussian chain elasticity

$$F_{\text{elastic}} / k_B T \approx R_e^2 / a^2 N$$

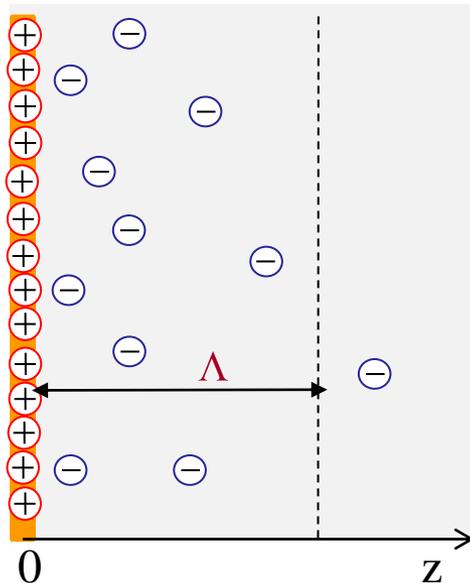
and intra-molecular\_Coulomb repulsion

$$F_{\text{Coulomb}} / k_B T \approx l_B Q^2 / R_e \quad \text{Bjerrum length } l_B = e^2 / \epsilon k_B T$$

Balancing  $F_{\text{Coulomb}} \approx F_{\text{elastic}}$  gives

$$R_e \approx a^{2/3} l_B^{1/3} N f^{2/3} \approx \xi_e N_b$$

number of blobs



Dimensionless  
 electrostatic potential

$$\psi(z) = e\Psi(z) / k_B T$$

Boundary conditions:

$$\left( \frac{d\psi}{dz} \right)_{z=0} = -\frac{2}{\Lambda} \quad \left( \frac{d\psi}{dz} \right)_{z=\infty} = 0$$

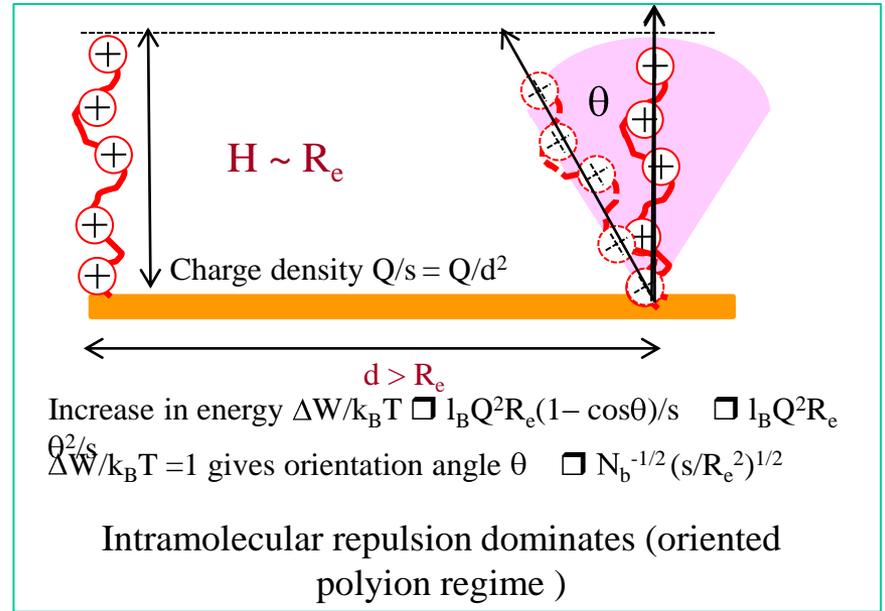
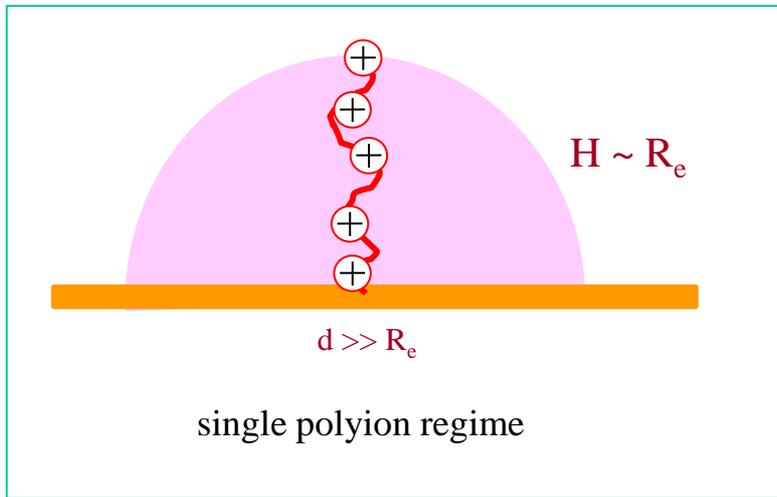
Gouy-Chapman length  $\Lambda = s / (2\pi Q l)$

Salt-free solution,  $c_+(z) = 0$

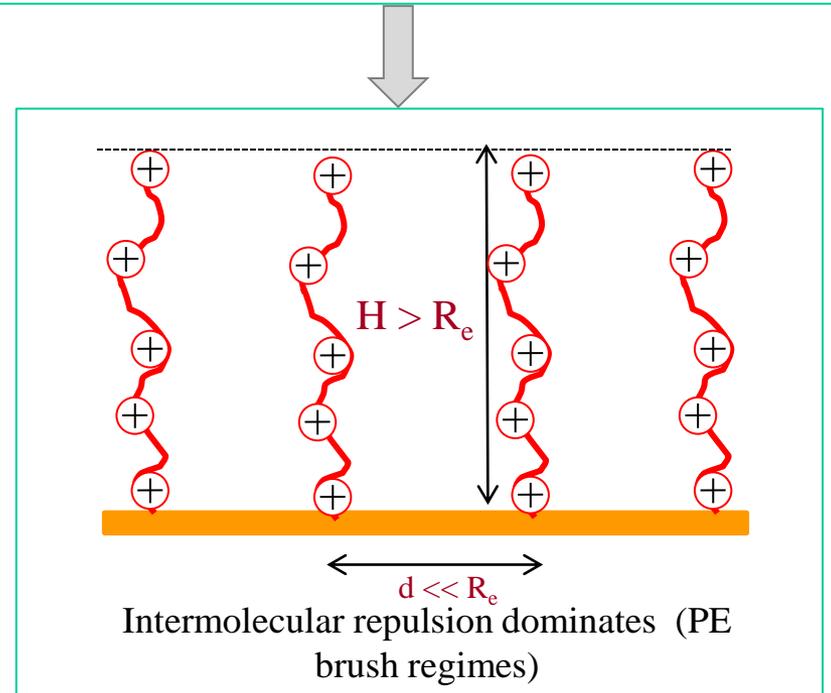
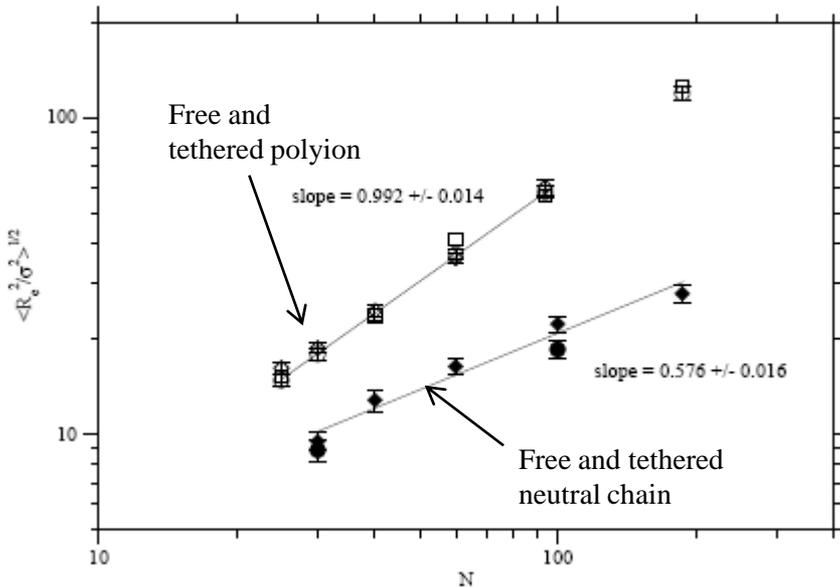
$$c_-(z) = 1 / [2\pi l_B (z + \Lambda)^2]$$

$$\psi(z) = -2 \ln[(z + \Lambda)]$$

# Tethering PE chains to substrate

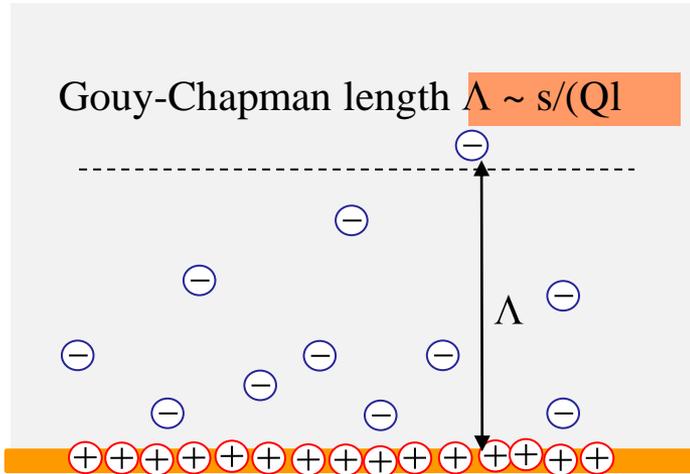


MD simulations, A.Kumar

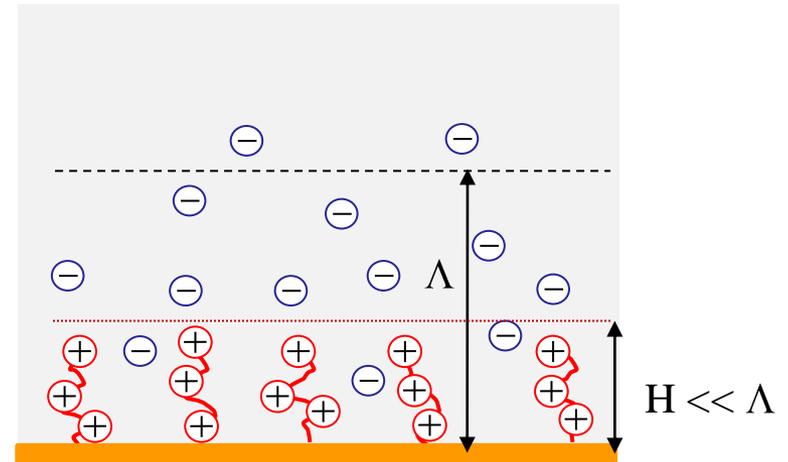


# Intermolecular repulsion regime

## Planar charged surface



Pincus, 1991



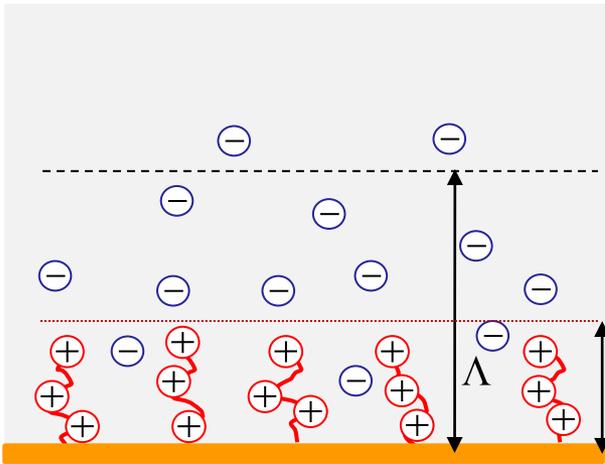
Majority of mobile counterions leave the brush, and chains are stretched due to intermolecular Coulomb repulsions

Stretching force  $F_{\text{electrostatic}} \sim k_B T l_B Q^2 / s$

Elastic force  $F_{\text{elastic}} \sim k_B T H / a^2 N$  (*Gaussian elasticity*)

Balance of forces gives  $H \sim a^2 l N^3 f^2 / s$ . Narrow regime

# Osmotic brush regime (low salt)

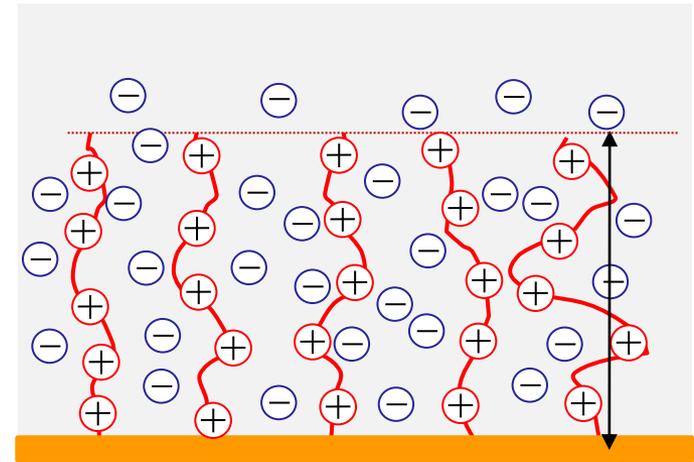


Increase in chain charge

$$Q = f N$$

or grafting density

$$\rho = d^{-2}$$



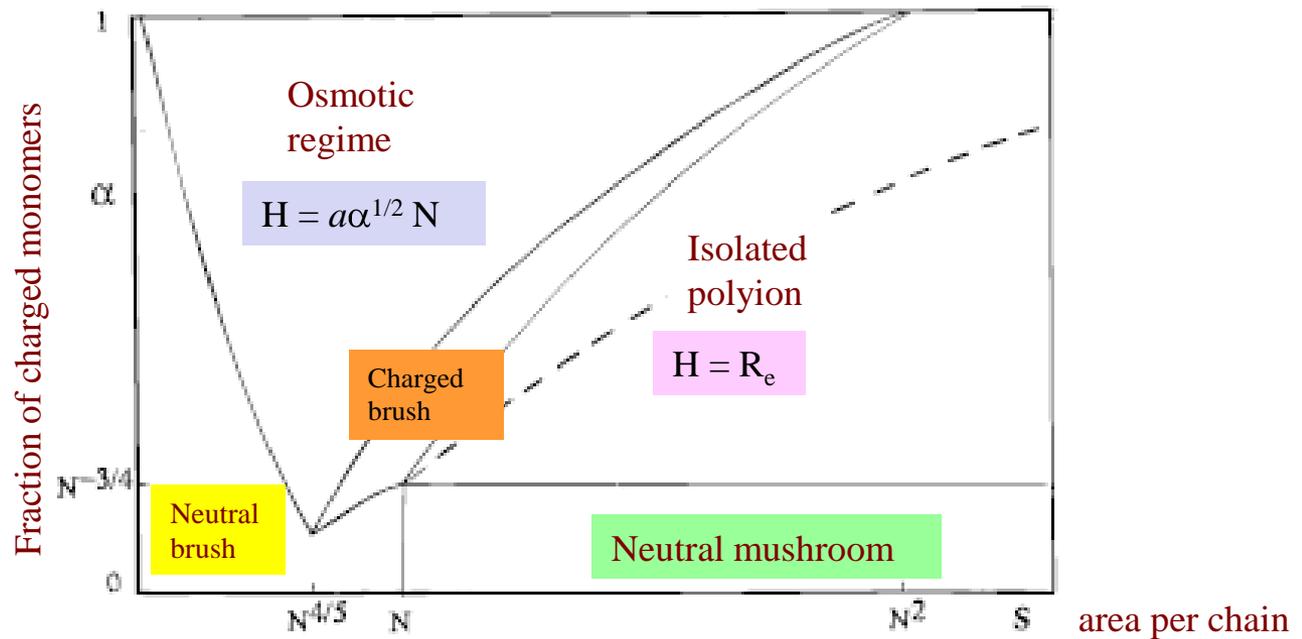
Majority of mobile counterions leave the brush

Strong attraction to polyions entraps counterions inside osmotic PE brush. Bare polyion charge is almost totally compensated.

Counterion osmotic stretching against Gaussian elastic force

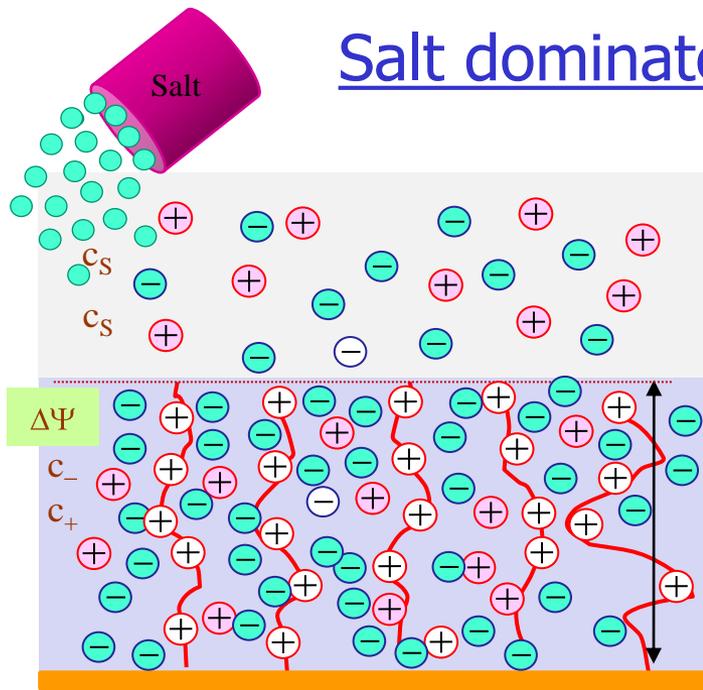
Balance of forces gives  $H \sim H_0 \sim a f^{1/2} N$ . Main regime of salt-free PE brush

# Diagram of states of PE brush (low salt)

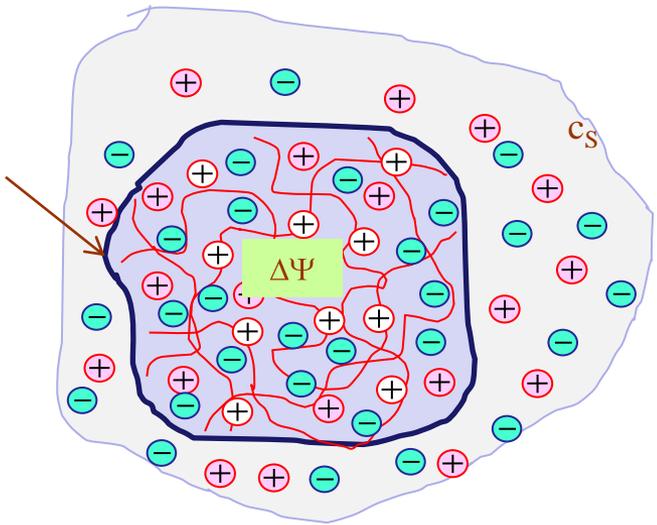


**Figure 2.** Diagram of states of a planar polyelectrolyte brush in  $\alpha, s$  coordinates.

# Salt dominated PE brush regime (high salt)



Volume with concentration of immobilized charge  $fc$



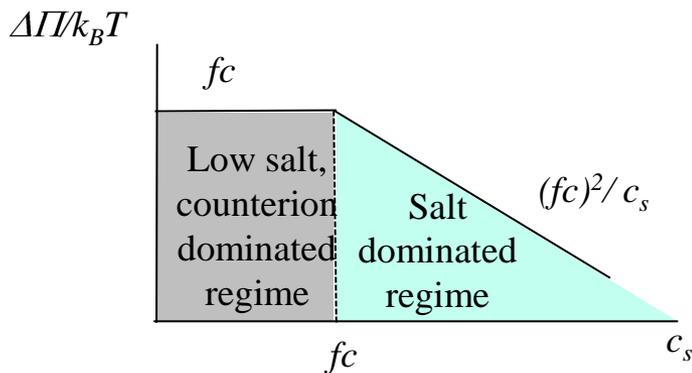
Electroneutrality:  $fc + c_+ = c_-$

Donnan equilibrium:  $c_+ c_- = c_s^2$

$c_+ = c_s \exp(-e\Delta\Psi/k_B T)$

$c_- = c_s \exp(e\Delta\Psi/k_B T)$

Differential ion pressure:  $\Delta\Pi/k_B T = c_+ + c_- - 2c_s$

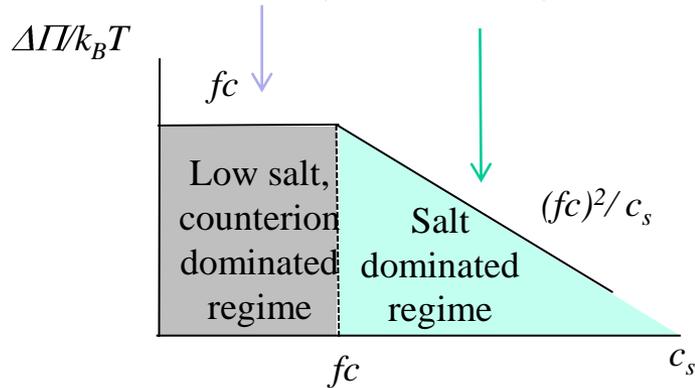


$$\simeq \begin{cases} fc & \text{when } fc \gg c_s \\ (fc)^2/c_s & \text{when } fc \ll c_s \end{cases}$$

# Effect of added salt : standard PE brush model

Elastic force  $F_{\text{elastic}} \sim k_B T H / a^2 N$  (Gaussian elasticity at all concentrations of salt  $c_s$ )

Two regimes for  $\Delta\Pi$ :  $fc \gg c_s$  and  $fc \ll c_s$

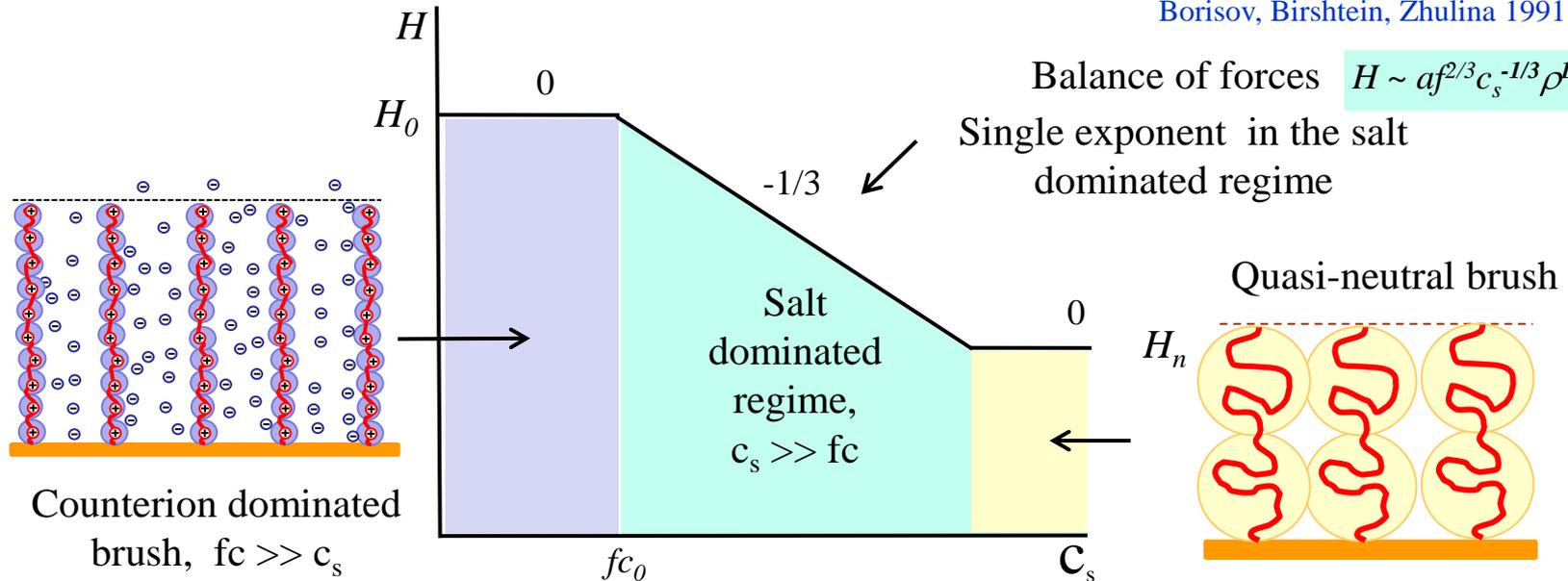


Pincus 1991

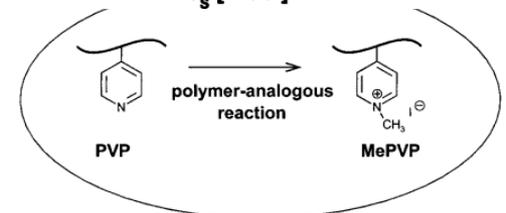
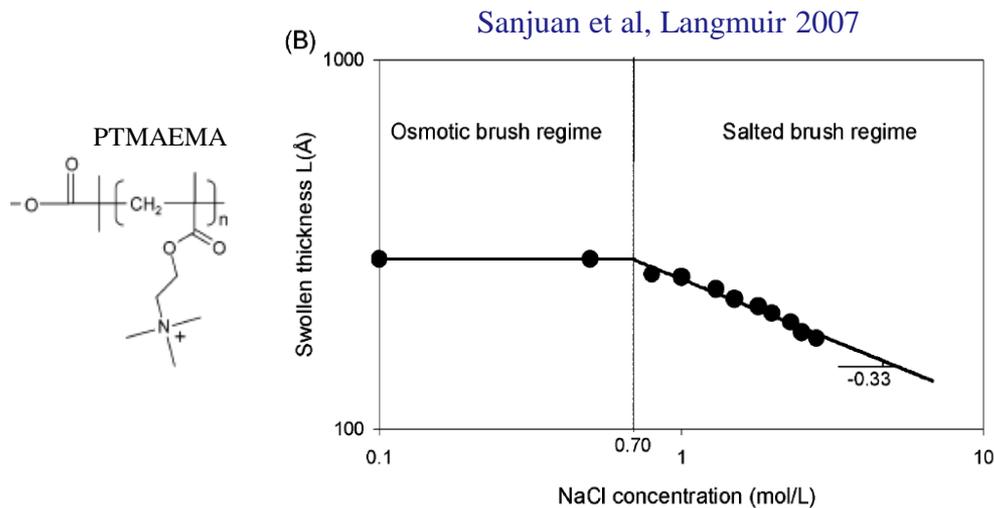
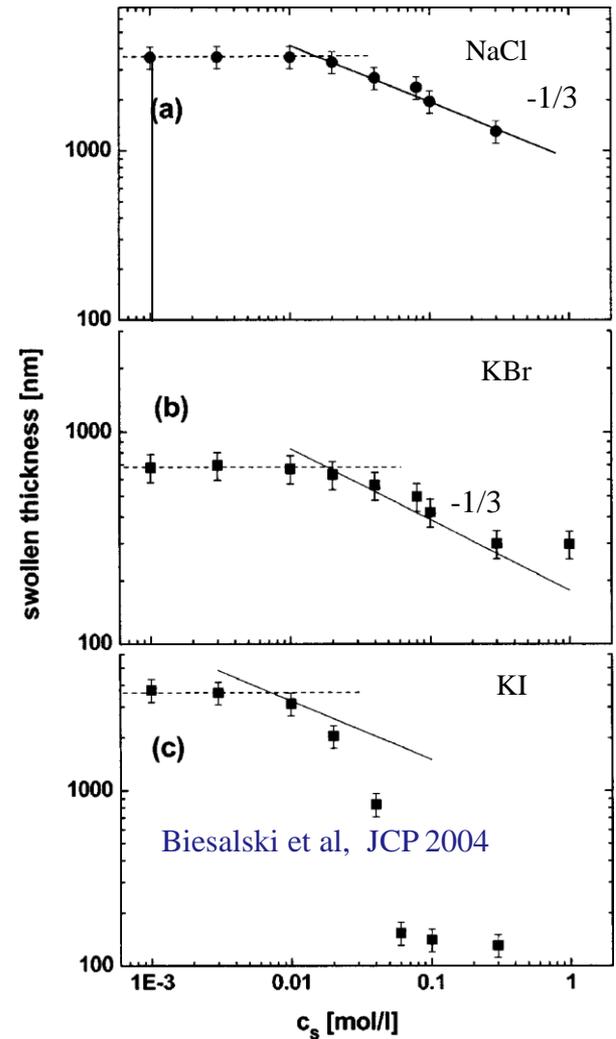
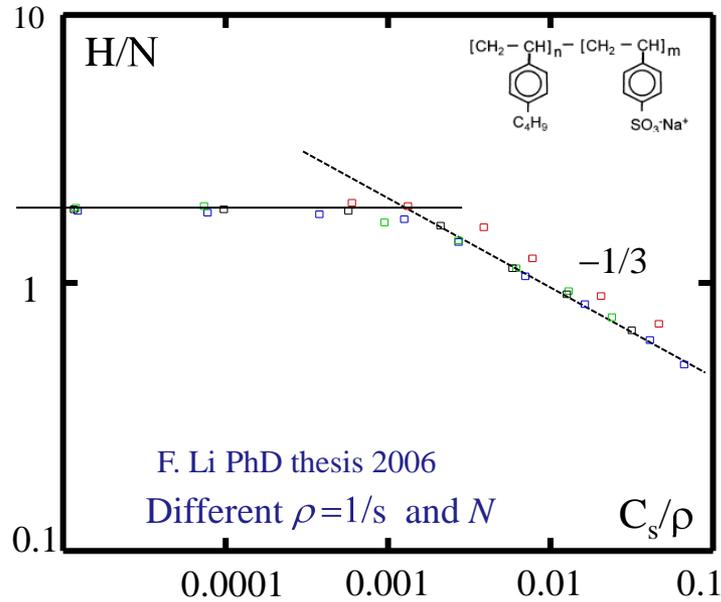
Borisov, Birshtein, Zhulina 1991

Balance of forces  $H \sim a f^{2/3} c_s^{-1/3} \rho^{1/3} N$

Single exponent in the salt dominated regime

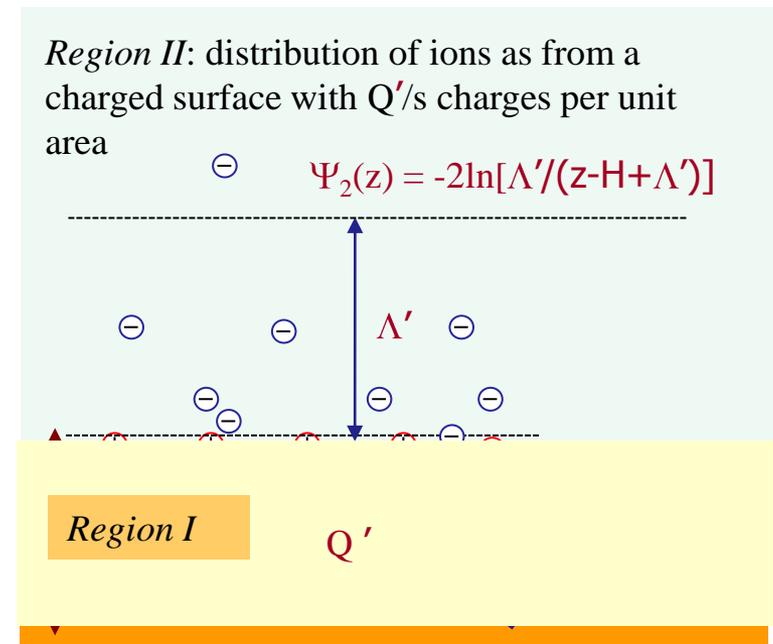
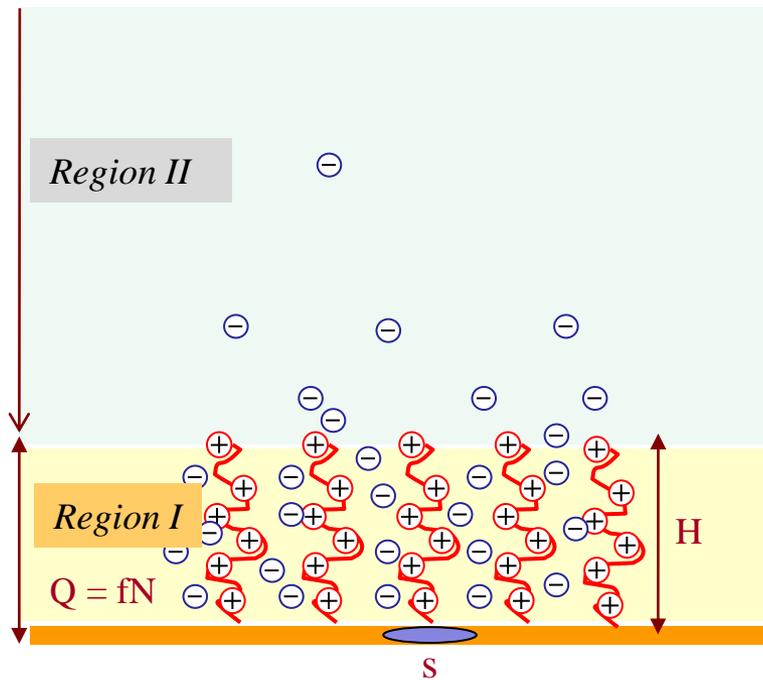


# Experimental test of osmotic and salt dominated regimes



# Self-Consistent Field (SCF) model of salt-free PE brush

Zhulina, Borisov 1997



Gouy-Chapman length  $\Lambda = s/(2\pi Q l_B)$  associated with net charge  $Q/s$  per unit area

Gouy-Chapman length  $\Lambda' = s/(2\pi Q' l_B)$  associated with uncompensated brush charge  $Q'/s$  per unit area

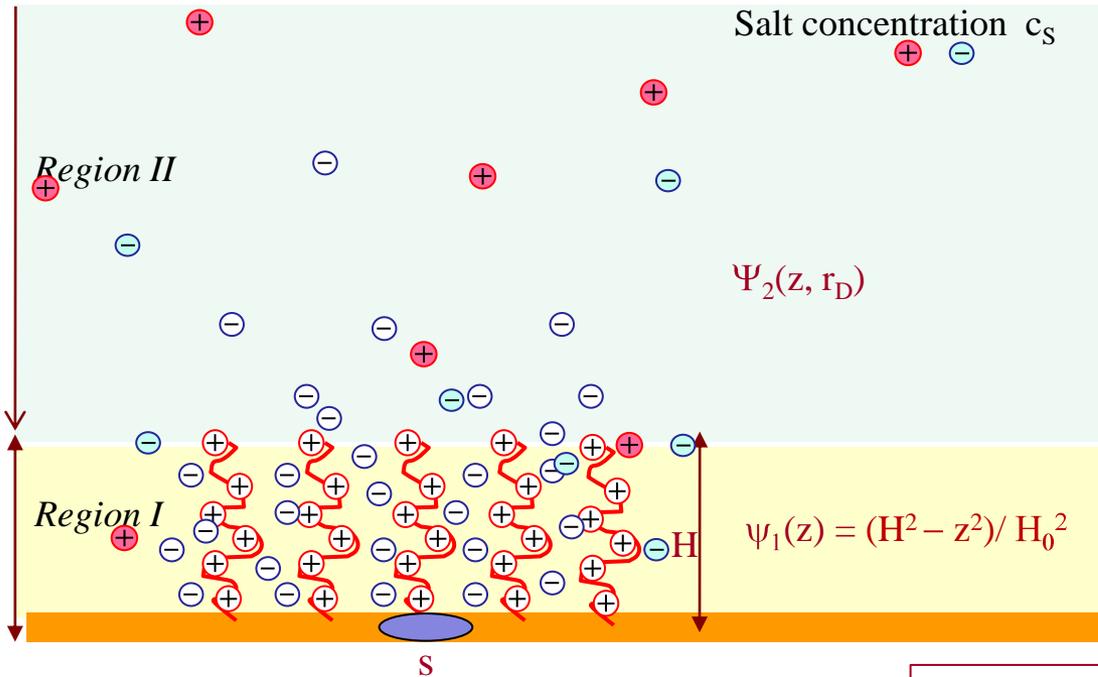
*Region I:* for chains with Gaussian elasticity and no excluded volume interactions, molecular field acting at a monomer located at distance  $z$  from the surface,  $\delta f_{\text{interaction}}[\phi]/\delta\phi = f\psi(z) = \text{Const} - 3\pi^2 x^2/8a^2 N^2$



Electrostatic potential in the brush  $\psi_1(z) = (H^2 - z^2)/H_0^2$  where  $H_0 = (8/3\pi^2)^{1/2} a f^{1/2} N$  is characteristic electrostatic length.

# SCF model of salt-added PE brush

Zhulina, Wolterink, Borisov 2000



## Important length scales:

electrostatic length  $H_0 = (8/3\pi^2)^{1/2} af^{1/2} N$

Debye radius  $r_D = (8\pi l_B c_s)^{-1/2}$

Gouy-Chapman length  $\Lambda = s/(2\pi Q l_B)$

## Reduced parameters:

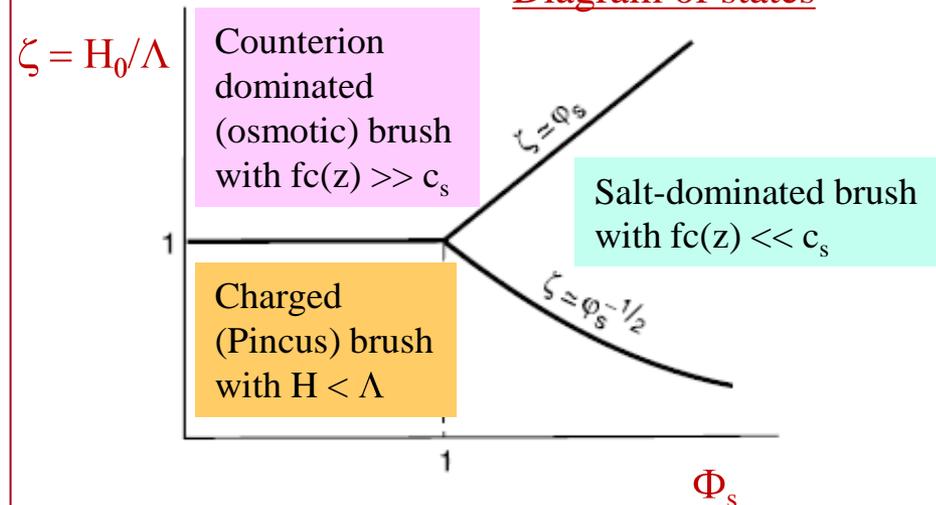
$\zeta = H_0/\Lambda$  (measure of PE brush charge compensation)

$h = H/H_0$

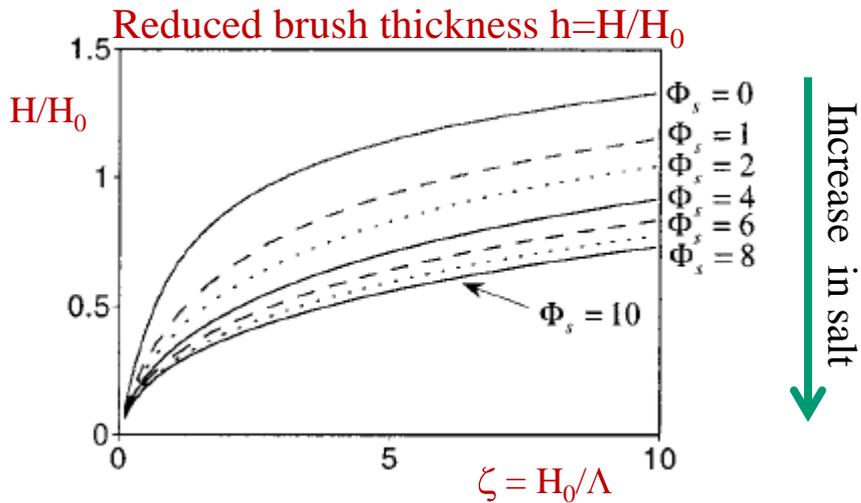
$\Phi_s = 2\pi l_B H_0^2 c_s = (H_0/2r_D)^2$

Output of SCF model:  
 confirmed scaling dependences  
 for PE brush thickness  $H$   
 + crossover regions  
 + internal brush structure  
 (density distributions).

## Diagram of states



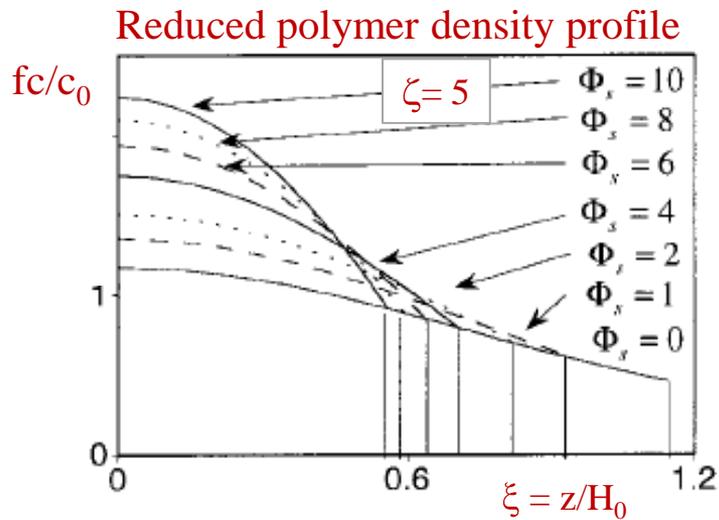
# Internal structure of PE brush



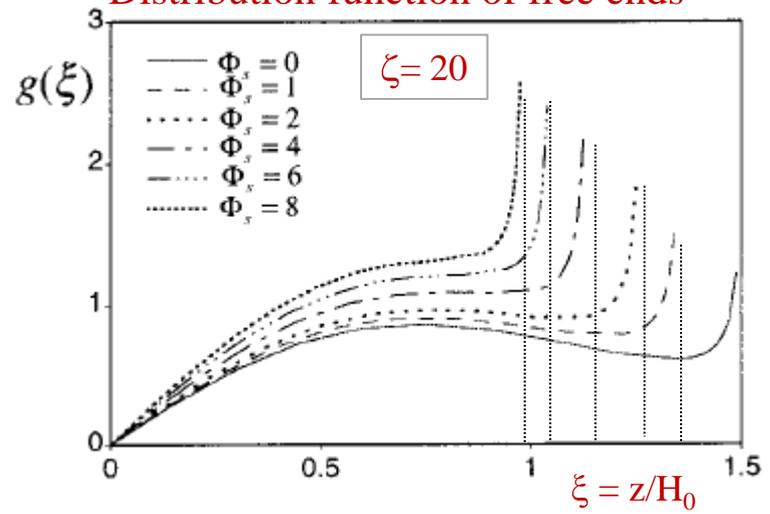
Reduced distance  $\xi = z/H_0$

Average net charge concentration

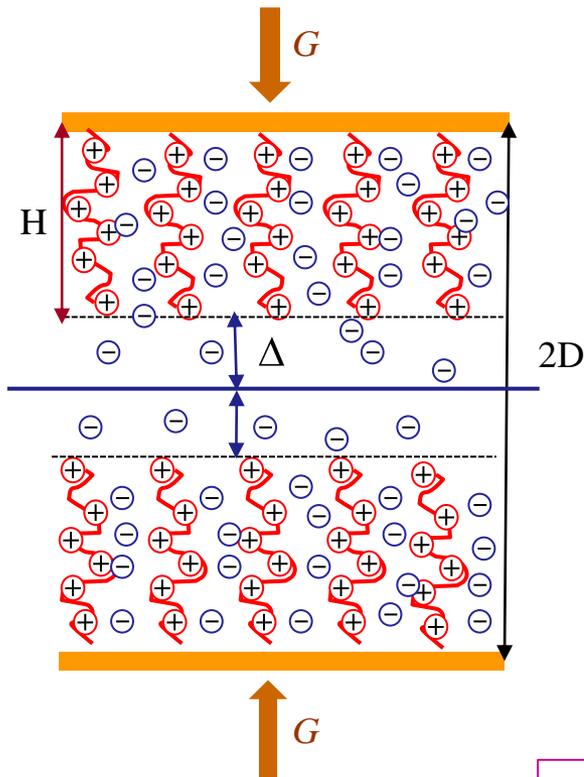
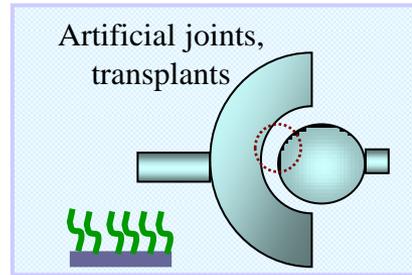
$$c_0 = fN/sH_0$$



Distribution function of free ends

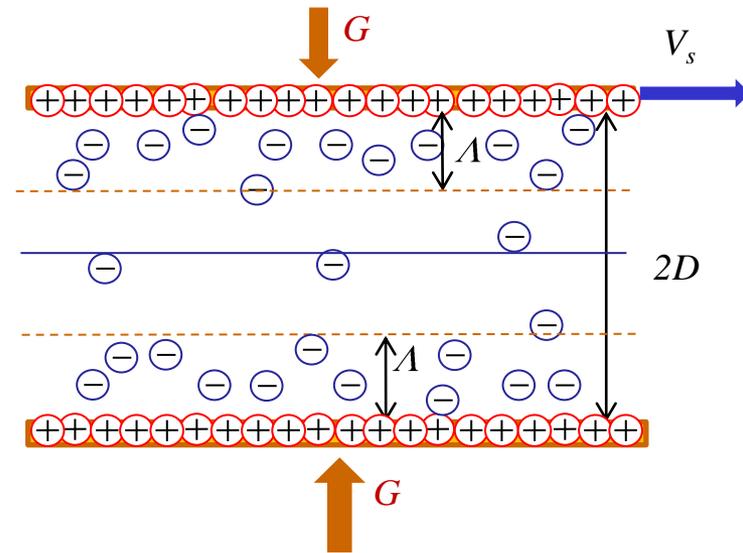


# Interactions between PE brushes



Normal force  $G$  per unit area (pressure)

For comparison: two surfaces with the same surface charge density  $Q/s$

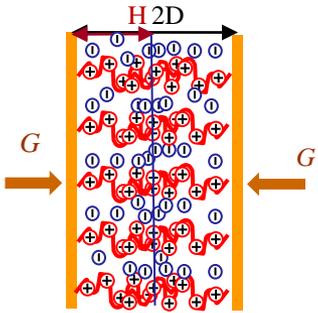


Goal: to highlight the role of tethered polymer on resistance to compression of two opposing PE bearing surfaces

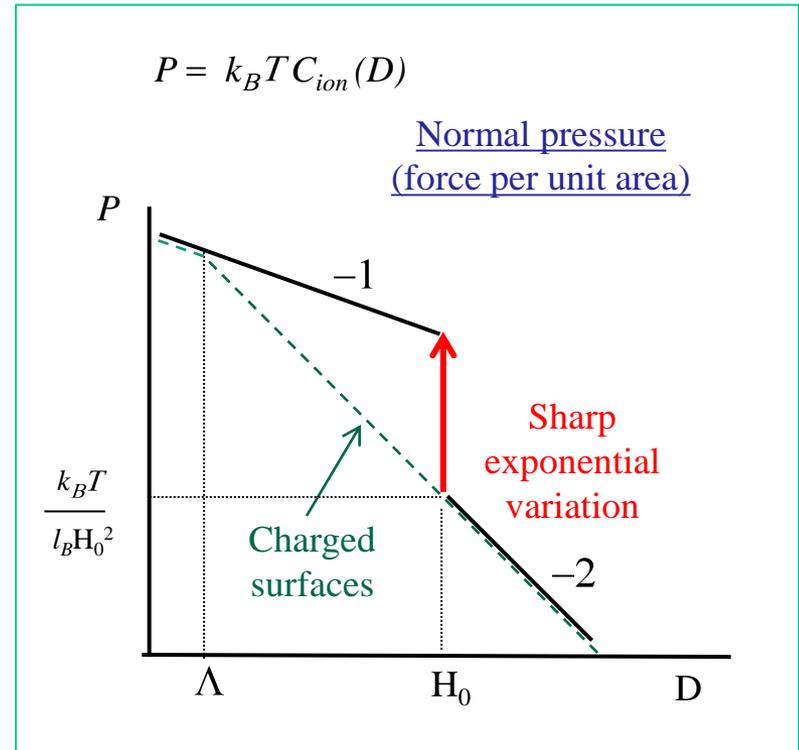
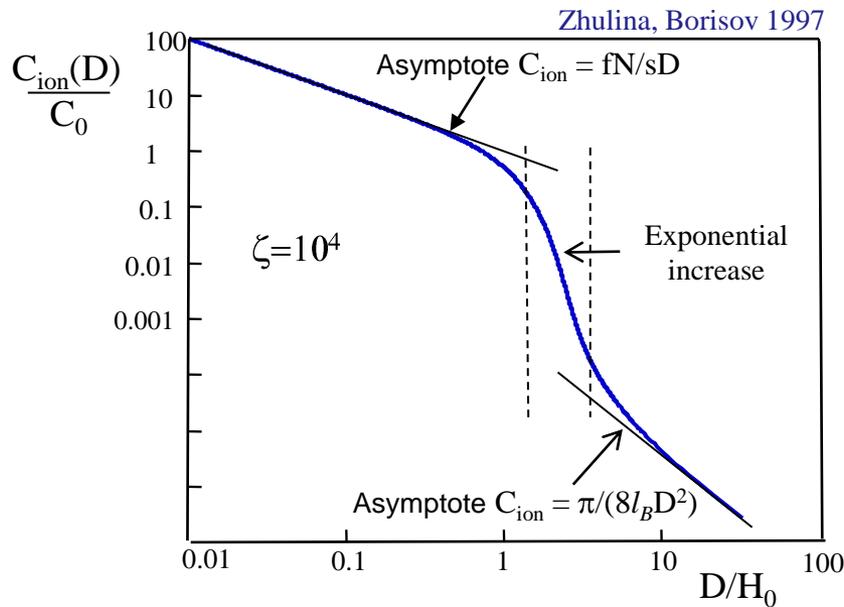
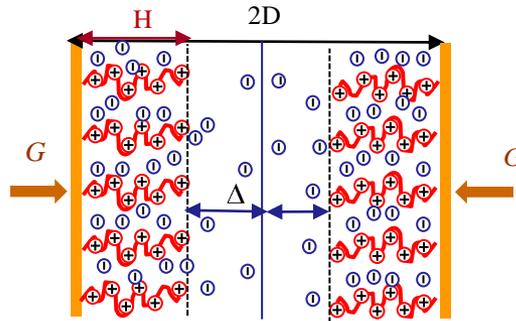


# Compression of apposing PE brushes *versus* planar surfaces

Strong compression



Weak compression



Two main regimes of PE brush compression:

$D \gg H_0$       $P \sim D^{-2}$  (compression of ion tail)

$D \ll H_0$       $P \sim D^{-1}$  (confinement of ions in the body of PE brush)

with **sharp crossover** between them

# Curved PE brushes (scaling model)

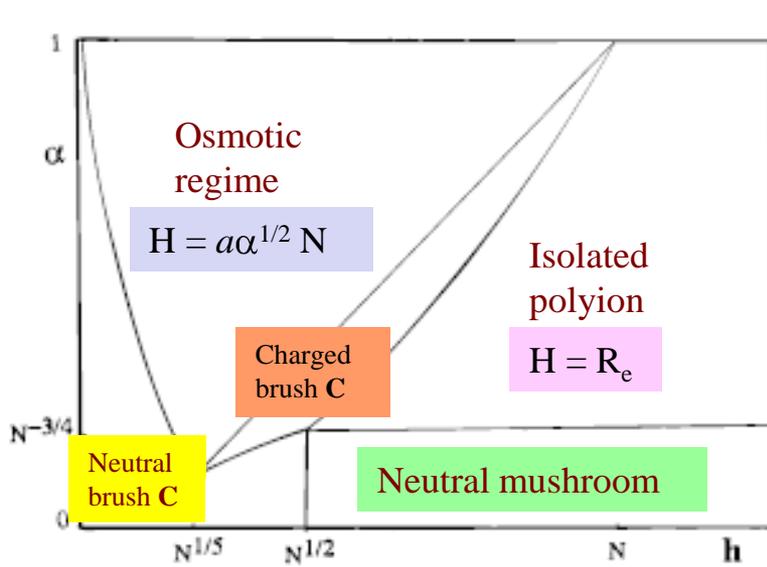
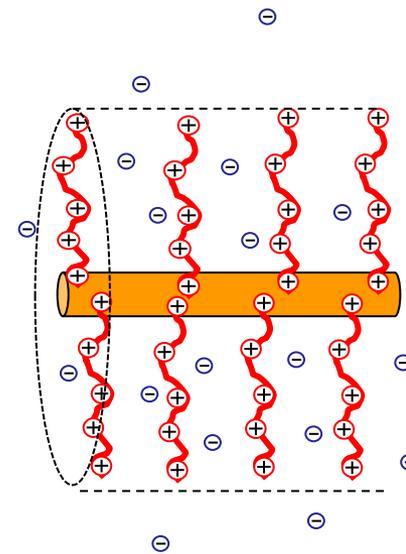


Figure 3. Diagram of states of a cylindrical (bottle) brush in  $\alpha, h$  coordinates.



Osmotic, isolated polyion and neutral mushroom regimes are similar for all geometries

Linear grafting density  $1/h$  chains

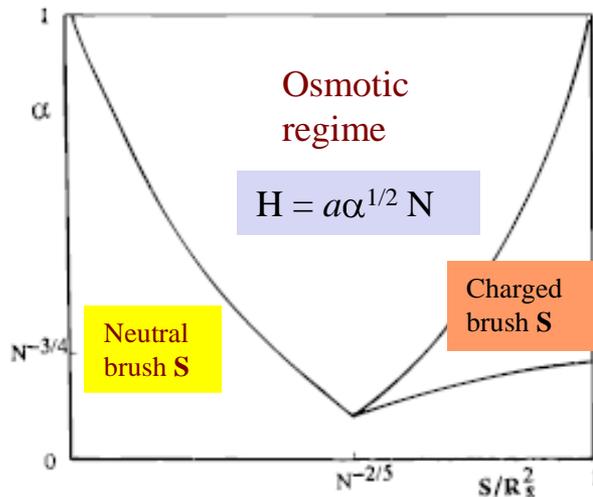
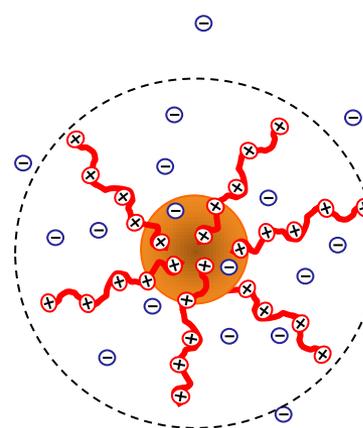


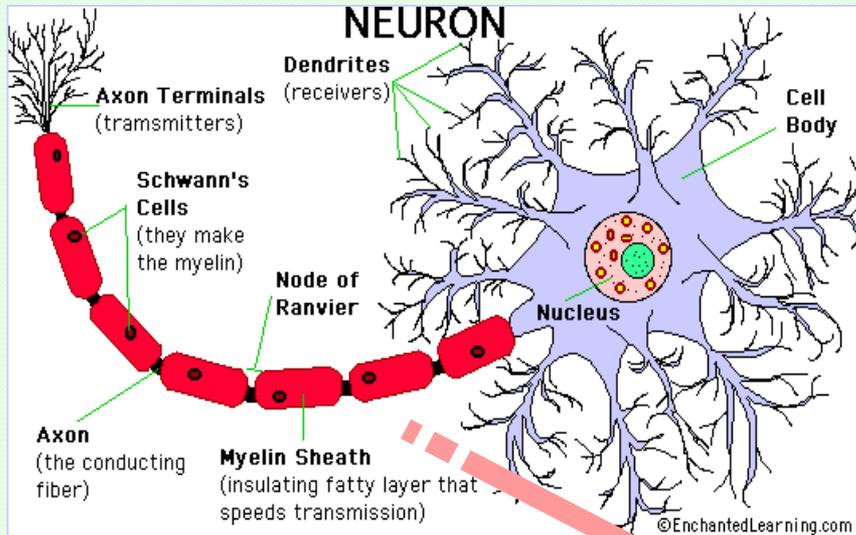
Figure 5. Diagram of states of a spherical brush in  $\alpha, s/R_s^2$  coordinates.



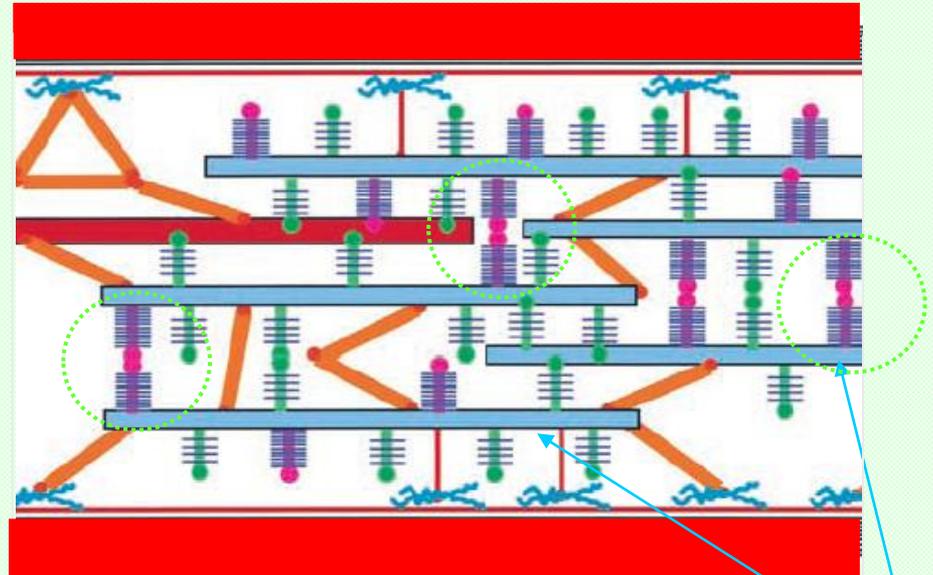
Scaling dependences in charged brush and neutral brush regimes depend on brush geometry

Angular grafting density  $p \gg 1$  chains

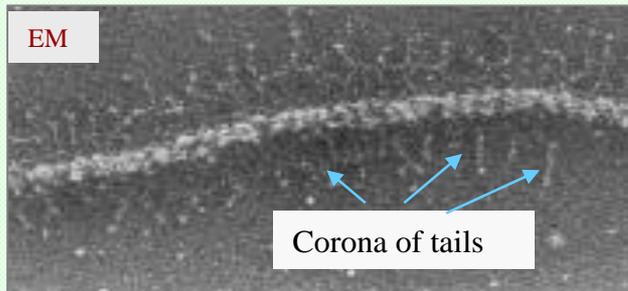
# Cytoskeleton in axon of neuron



Schematic of axonal cross-section Rao et al *JCB* 2002

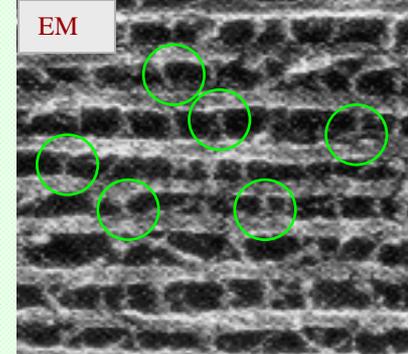


Individual neurofilament NF



Fuchs & Cleveland *Science* 1998

Hirokawa *JCB* 2000



Neurofilament network in axon



## Features of NF brush

- Cylindrical charged brush (with thickness  $D \gg$  core radius  $R$ )
- Combines weak and strong polyelectrolyte properties

- Complex architecture:

Three types of projections, **H**- **M**- and **L**- tails, with different lengths and degrees of ionization, NF composition varies;

Phosphorylation changes charge distribution on the longest **H**-tail: transforms gradient polyampholyte into block polyelectrolyte;

Shortest **L**-tails are necessary for filament assembly (they must be present in NF brush)

- Long tails can cross-bridge

# SF-SCF modeling NF brush of **H**, **M** and **L** tails with amino acid resolution

Coarse-grained tails with **no** elements of secondary structure, all monomers have same size  $a = 0.6 \text{ nm}$

Conservation of actual charge distributions on the tails (+ phosphorylation of serines in **KSP** motifs)

Tethering tails with varied ratio (**L:M:H** = 7:3:2 in wild-type NF) to impermeable charged cylinder that mimics **NF** backbone

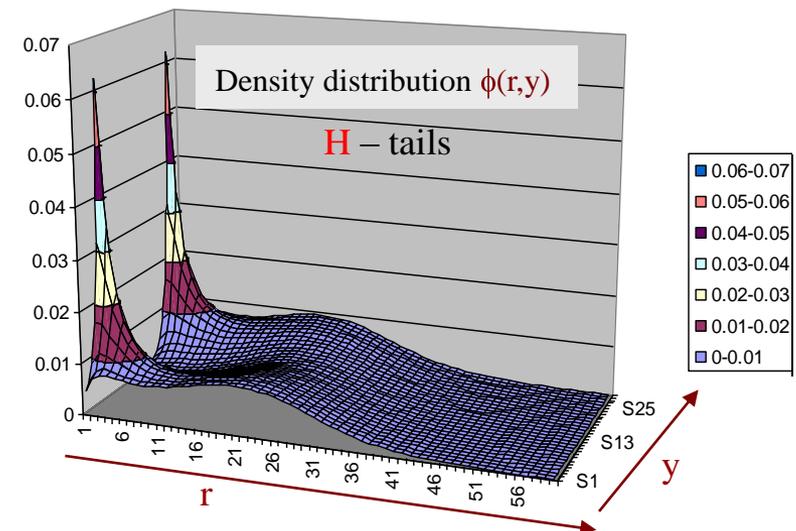
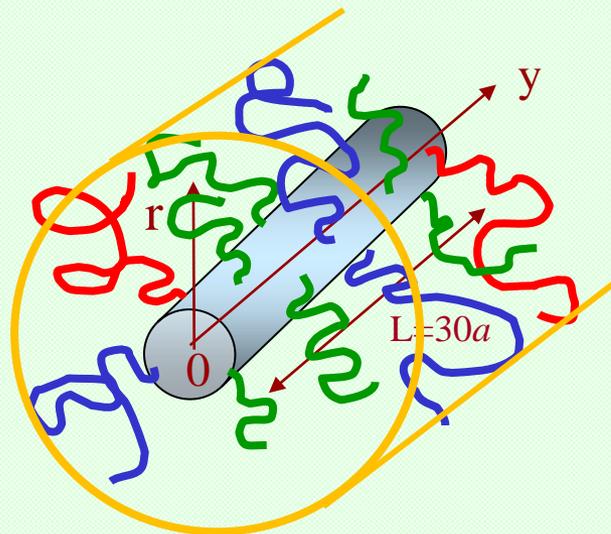
## SF-SCF numerical method:

Electrostatic interactions are treated on the Poisson-Boltzmann level

Nonelectrostatic interactions are treated within the Bragg-Williams approximation

Gaussian statistics for tethered chains

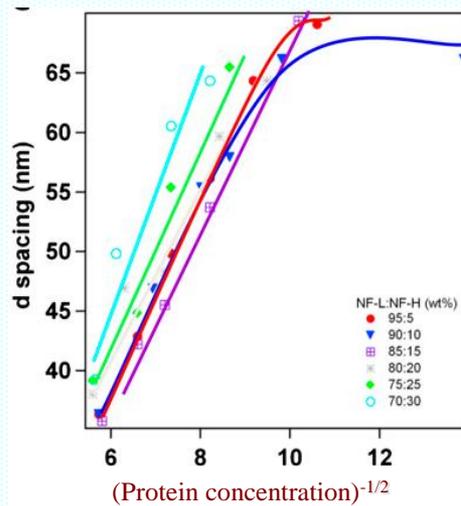
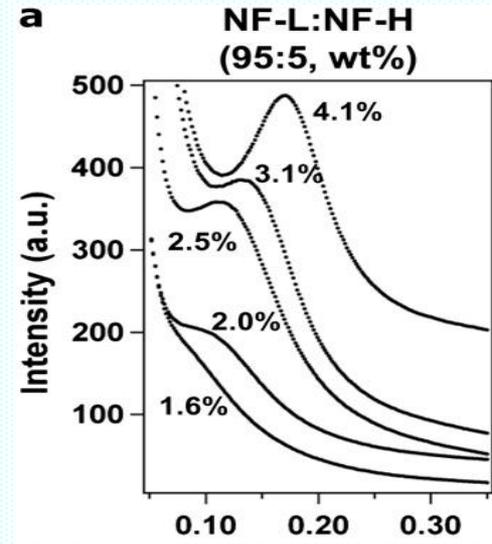
1-G, 2-G versions



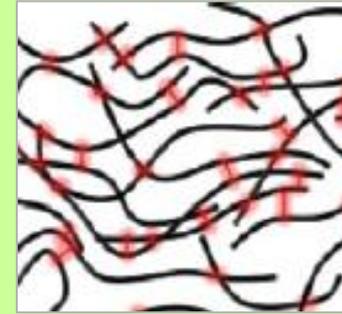
# NF-NF interactions: nematic order in neurofilament hydrogel

J.Jones & C.Safinya *BJ* 2008

Nematic-to-isotropic transition in solution of reconstituted LH-filament (SAXS)

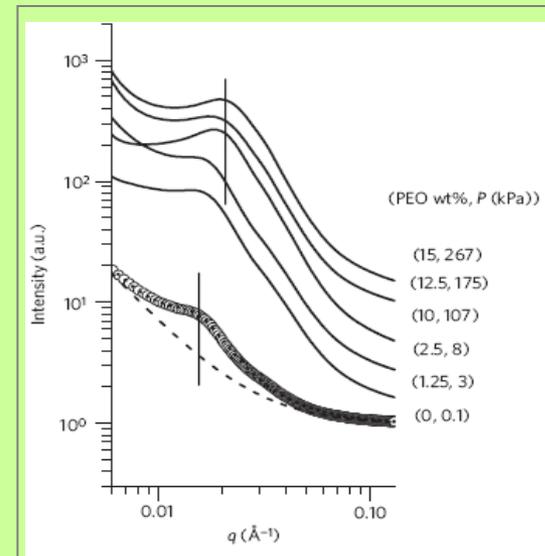


NF-NF distance  $d \sim \phi^{-1/2}$



Reconstituted NFs of various compositions (ratios **H:M:L**) subjected to external pressure **P** due to added **PEG** (plotted along Y-axis)

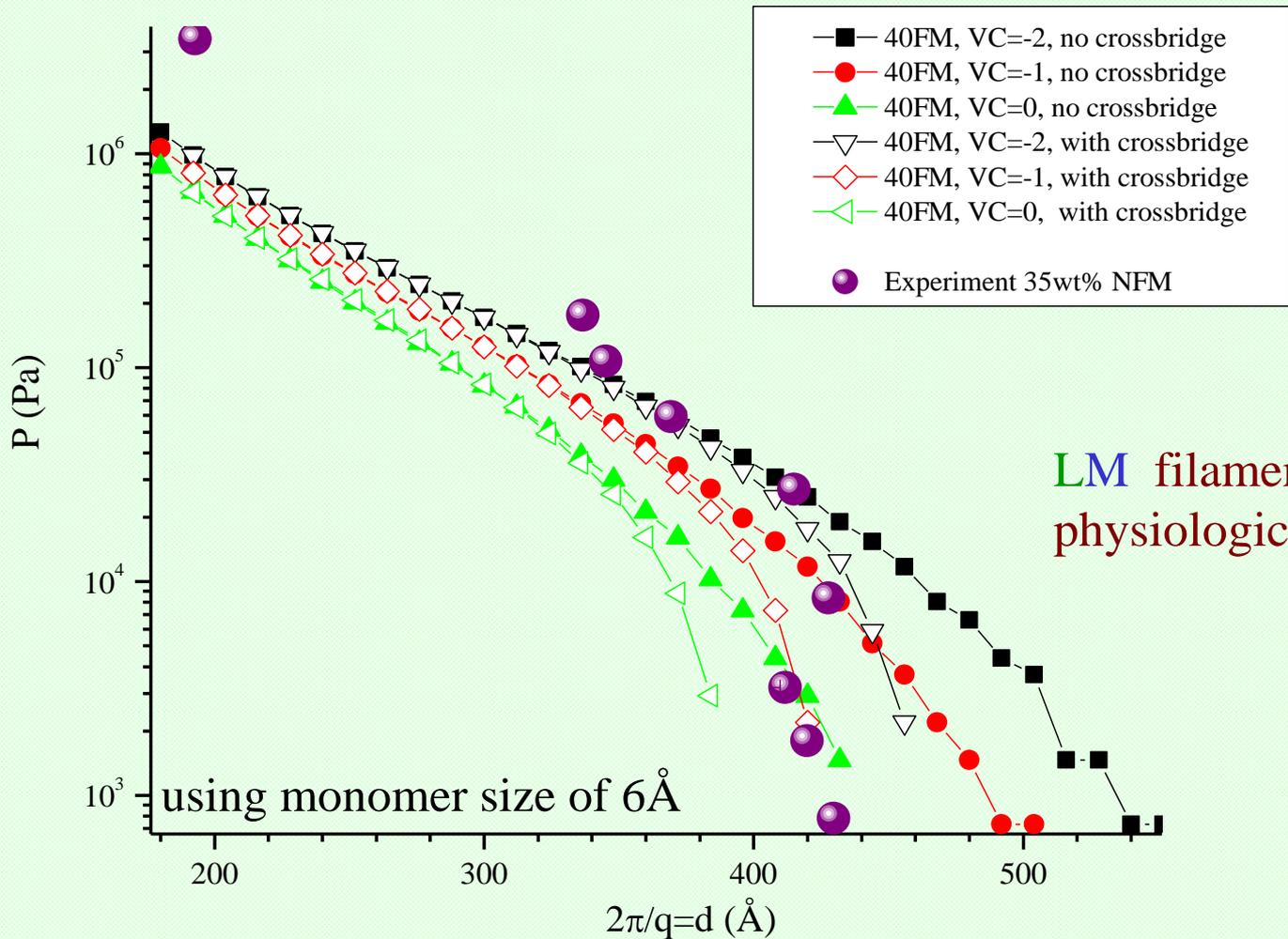
R.Beck et al. *Nature Materials* 2010



# Comparison between SF-SCF and experimental pressure-distance profiles

Theory vs. Experiment: 150mM

$$P_{\text{theor}} = -d(F^{\text{int}})/dV$$



LM filament under physiological conditions

# Comparison between SF-SCF and experimental pressure-distance profiles

Wild-type neurofilament  
 LMH 7-3-2, 150mM

