Nonequilibrium work relations III. Dissipation and the Arrow of Time

second law of thermodynamics (macroscopic) $\rightarrow W \ge \Delta F$ nonequilibrium work relations (microscopic) $\rightarrow \langle e^{-\beta W} \rangle = e^{-\beta \Delta F}$, etc.

What is the precise relationship? What does the 2nd law "look like" at the microscale?

Inequalities:

Jensen's inequality
$$\rightarrow \langle e^x \rangle \ge e^{\langle x \rangle}$$

holds for any real, convex function
of a real variable
Simple derivation of Jensen's inequality:
convex function $A(x)$, $A''(x) \ge 0 \quad \forall x$
probability distribution
 $p(x)$, $1 = \int dx \, p(x)$, $\bar{x} = \int dx \, x \, p(x)$
define a new function $B(x) = A(\bar{x}) + (x - \bar{x})A'(\bar{x})$
(tangent to $A(x)$)
by construction: $A(x) \ge B(x) \quad \forall x$
 $\langle A(x) \rangle \ge \langle B(x) \rangle = \int dx \, p(x) [A(\bar{x}) + (x - \bar{x})A'(\bar{x})] = A(\bar{x})$



now let's obtain a somewhat stronger result ...



Guessing the direction of Time's Arrrow:

In forward/reverse processes, trajectories come in conjugate pairs: if X denotes a possible realization of the forward process, then its conjugate twin X⁺ is a possible realization of the reverse process.

Suppose you are shown a movie depicting the microscopic evolution of the system as $\lambda : A \rightarrow B$ (forward process). How can you tell whether you are viewing (1) the events in the order in which they actually occurred, or (2) a movie of the reverse process, run backward ?

exercise in statistical inference:

given the observed data, which hypothesis is more likely?

Bayes' theorem

$$L(hyp \mid dat) \propto P(dat \mid hyp) \cdot P_0(hyp)$$

e.g. $L(F \mid X) \propto P^F(X) \cdot P_0(F)$

assume equal priors : $P_0(F) = P_0(R) = 1/2$

$$L(F \mid X) = \frac{P^F(X)}{P^F(X) + P^R(X^+)} = \frac{1}{1 + \left[P^R(X^+)/P^F(X)\right]}$$
$$= \frac{1}{1 + e^{-\beta(W - \Delta F)}}$$

[Shi03,Mar07]



- when W > ΔF, it is more likely that we are seeing the events in the correct order ("forward"), while for W < ΔF it is the other way around
- the transition from "almost certainly reverse" (L≈0) to "almost certainly forward" (L≈1) happens over a few k_BT (consistent w/ earlier results: very low probability to see second law "violated" by more than a few k_BT)

Relative entropy and dissipation:

Given two normalized probability distributions p(x) & q(x), the *relative entropy* of p with respect to q is

$$D(p \mid q) = \int dx \, p(x) \ln \frac{p(x)}{q(x)} \ge 0$$

(aka *Kullback-Leibler divergence*)... provides a measure of the degree to which one distribution is distinguishable from the other

Let's use this to quantify thermodynamic irreversibility.

 $P^{F}(X)$ = distribution of forward trajectories $P^{R}(X^{+})$ = distribution of reverse trajectories

The relative entropy between these two distributions measures *time-reversal asymmetry*. (How differently does the system respond in the two processes?)

$$D(P^{F} | P^{R}) = \int dX P^{F}(X) \ln \frac{P^{F}(X)}{P^{R}(X^{+})}$$
$$= \int dX P^{F}(X) \beta \left[W^{F}(X) - \Delta F \right]$$
$$= \beta \left(\left\langle W \right\rangle^{F} - \Delta F \right) = \beta W_{diss}^{F}$$

This result relates a *physical* measure of irreversibility (dissipated work) to an *information-theoretic* measure of time-reversal asymmetry (relative entropy).

Consistent w/ macroscopic experience:

$$W_{diss} >> k_B T$$
 , $D >> 1$ 33

References

Below is a short list of papers cited in these lecture notes. For a more comprehensive reference list, see the bibliography in my mini-review in *Eur. Phys. J. B* **64**, 331-340 (2008)

[Boc77] G.N. Bochkov and Y.E. Kuzovlev, Sov Phys JETP 45, 125 (1977) [Boc81] G.N. Bochkov and Y.E. Kuzovlev, Physica 106, 443 (1981) & *Physica* **106**, 480 (1981) [Cro98] G.E. Crooks, J Stat Phys 90, 1481-1487 (1998) [Cro99] G.E. Crooks, *Phys Rev E* 60, 2721-2726 (1999) [Gas04] P. Gaspard, J Stat Phys 117, 599-615 (2004) [Har07] N.C. Harris, Y. Song, C.-H. Kiang, Phys Rev Lett 99, 068101 (2007) [Hum01] G. Hummer and A Szabo, Proc Natl Acad Sci 98, 3658-3661 (2001) [Hum05] G. Hummer and A Szabo, Acc Chem Res 38, 504-513 (2005) [Jar97a] C. Jarzynski, *Phys Rev Lett* **78**, 2690-2693 (1997) [Jar97b] C. Jarzynski, *Phys Rev E* 56, 5018-5035 (1997) [Jar01] C. Jarzynski, Proc Natl Acad Sci 98, 3636-3638 (2001) [Jar02] C. Jarzynski, *Phys Rev E* 65, 046122 (2002) [Jar04] C. Jarzynski, J Stat Mech P09005 (2004) [Jar06] C. Jarzynski, *Phys Rev E* **73**, 046105 (2006) [Kaw07] R. Kawai, J.M.R. Parrondo, C. Van den Broeck, *Phys Rev Lett* **98**, 080602 (2007) [Lip02] J. Liphardt et al, Science 296, 1832-1835 (2002) [Mae99] C. Maes, J Stat Phys 95, 367-392 (1999) [Mar07] P. Maragakis et al, J Chem Phys 129, 024102 (2008) [Obe05] H. Oberhofer, C. Dellago, P.L. Geissler, J Phys Chem B **109**, 6902-6915 (2005) [Sei05] U. Seifert, Phys Rev Lett 96, 040602 (2005) [Shi03] M.R. Shirts et al, *Phys Rev Lett* **91**, 140601 (2003)