

introductory words.

Glasses as materials - polymer glasses/composites, optical fiber (scattering section very small due to homogeneity making possible to transmit over 1000 kilometers without amplification)

Dramatic slowdown with temperature between 10 and 19 orders of magnitude - 10^9 - 10^{19} and over.

Different types of interactions: Van der Waals, dipole-dipole, hydrogen bonding, metallic bonding.

↳ Crystallization? Problem for metallic glass.

$$VTF \quad \log \frac{C}{C_0} = A + \frac{B}{T-T_g} \rightarrow \text{Super Arrhenius behavior.}$$

Little experiment in office: $T_m = 30K$ $T_g \approx \text{Instantaneous}$
tube of metallic system 10^{21} molecules $20 \text{ years} \approx 10^{48} \text{ s}$ ↳ if critical nucleus ≈ 100 molecules
 $\left(\frac{10^{21}}{10^2}\right) \left(\frac{10^3}{10^{-6}}\right) \approx 10^{37}$ chances to nucleate
 possible $\xrightarrow{\text{time}} \xrightarrow{\text{nucleus}} \xrightarrow{\text{# of attempts}}$ still a liquid,
 fragile times longer super stable.

fragile / strong: fragile = with respect to perturbation caused by temperature

$$\text{def: kinetic fragility } m = \left| \frac{d \log \tau}{d(T_g/T)} \right|_{T_g} \quad \text{SiO}_2 \quad m \approx 16 \rightarrow \text{strong limit}$$

highest value ≈ 180 - → fragile limit.

"single time scale": Dielectric relaxation - collective measurement of dipole orientation

$$\begin{aligned} \text{Correlation function } CF(t) &\sim \langle (\sum_i \vec{\mu}_i(t)) \cdot (\sum_i \vec{\mu}_i(0)) \rangle \text{ starting times} \\ \text{NMR} \quad CF(t) &\sim \langle P_2 \vec{q}(t) \vec{q}(0) \rangle \text{ starting times, molecules} \end{aligned}$$

↳ liquids - Dielectric relaxation
Heat capacity spectroscopy ↳ input power (ω)
measure: $T(\omega)$

→ Eastwood et al J Phys Chem B 2013

* IMPORTANT EXCEPTION: translational diffusion coefficient → does not depend as much on temperature -

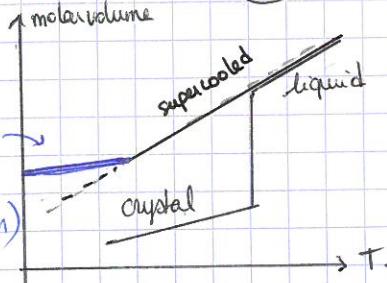


probe after heating and cooling how much the interface blurs ⇔ how much diffusion took place in the sample -

$$\frac{(D\alpha)_{H,T}}{(D\alpha)_{EOT}} = 100 = \frac{(D\alpha)_{air}}{(D\alpha)_{EOT}} \quad \text{molecular reorientation time}$$

thermodynamics -

glasses, gets stuck in conf with too much volume (depends on the preparation)



Convention for T_g : → cool at 10 K/min
→ $T_g = 100^\circ\text{C}$ at equilibrium,

(LC)

... O, monoatomic molecules - supercooled liquid (SCL) → liquid crystal → Glass of LC

Kauzmann entropy crises:

the excess entropy is $S_{ex} = S_{Scl} - S_{CR}$

what we want is the configurational entropy, that can be estimated as: $S_c \approx S_{Scl} - S_{Gibbs}$

some words about structure:

→ changes in S_{Gibbs} with respect to T are small

↔ connection between structure and dynamics

connections between dynamics and thermodynamics?

